



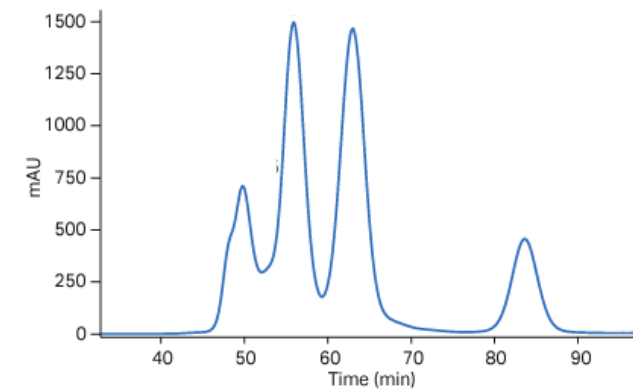
UNIVERSITY OF
CAMBRIDGE

Protein Purification Part 3

Day 5: Friday 24th March

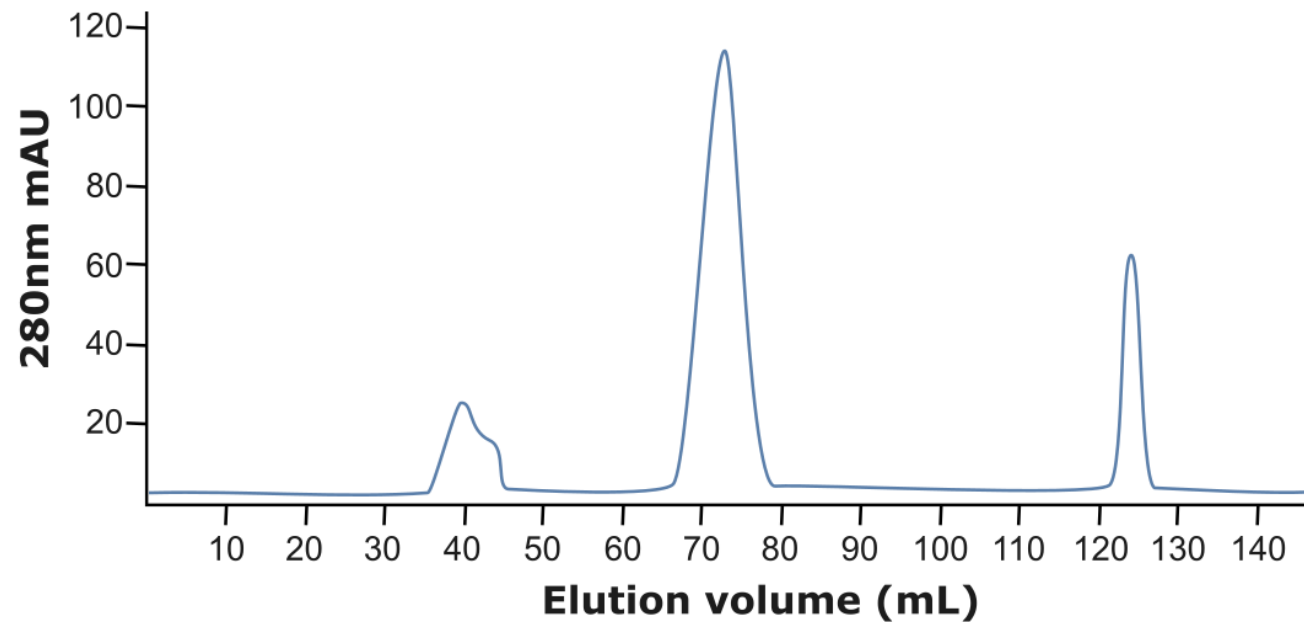
Continuing Protein Purification – Quality Control

- In Parts 1 and 2 we learnt about:
 - Chromatography techniques
 - Chromatography equipment
- Today we'll learn about:
 - Interpreting your data
 - Checking your protein quality
 - Checking for correct protein folding
 - How to store your protein
 - Troubleshooting difficult samples



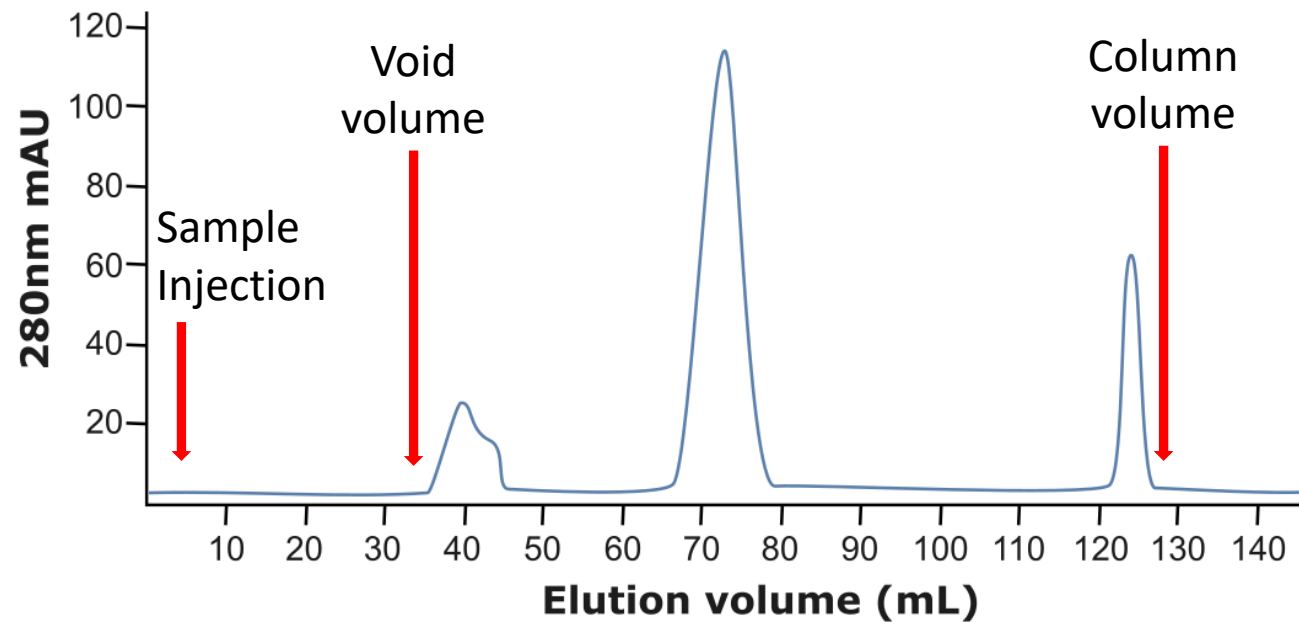
Interpreting Your SEC Elution Profile

- What can the elution volume tell us?

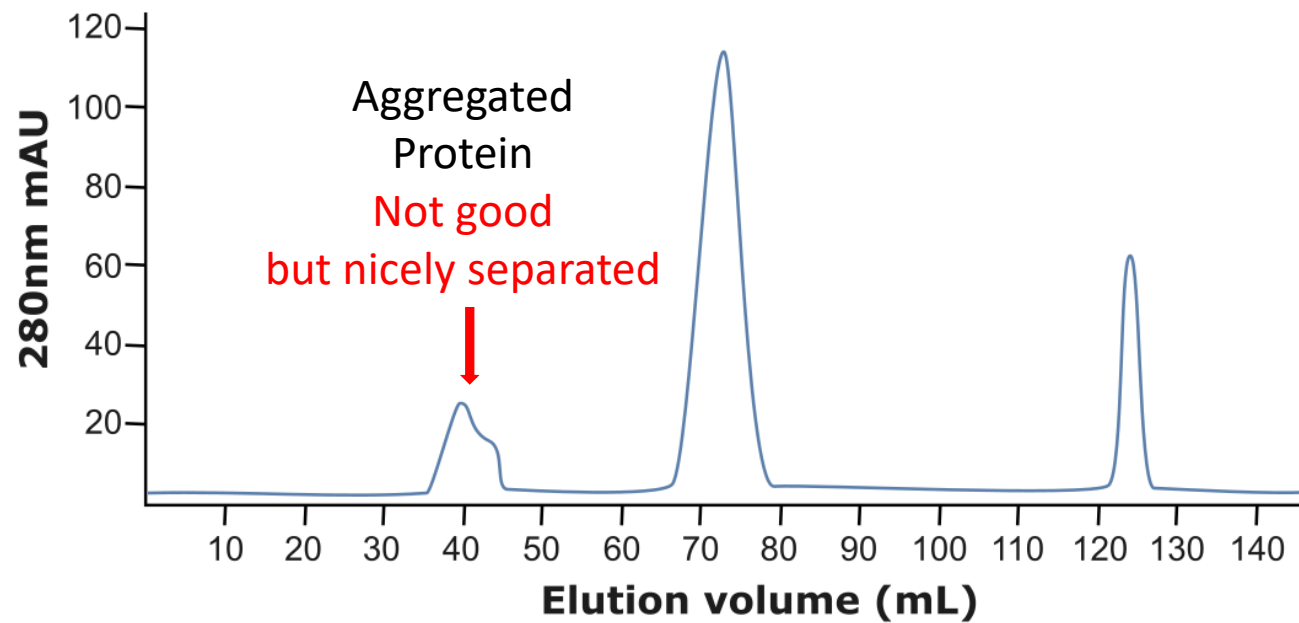


Interpreting Your SEC Elution Profile

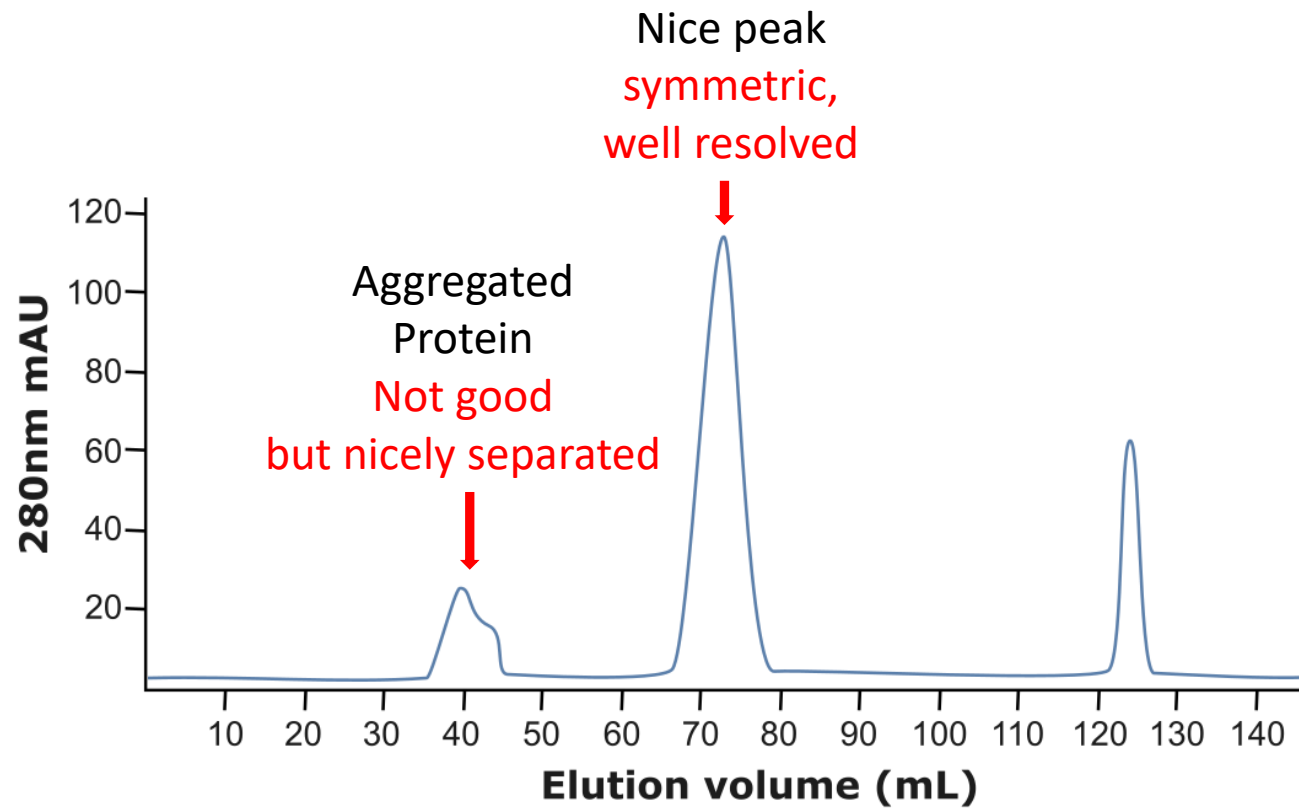
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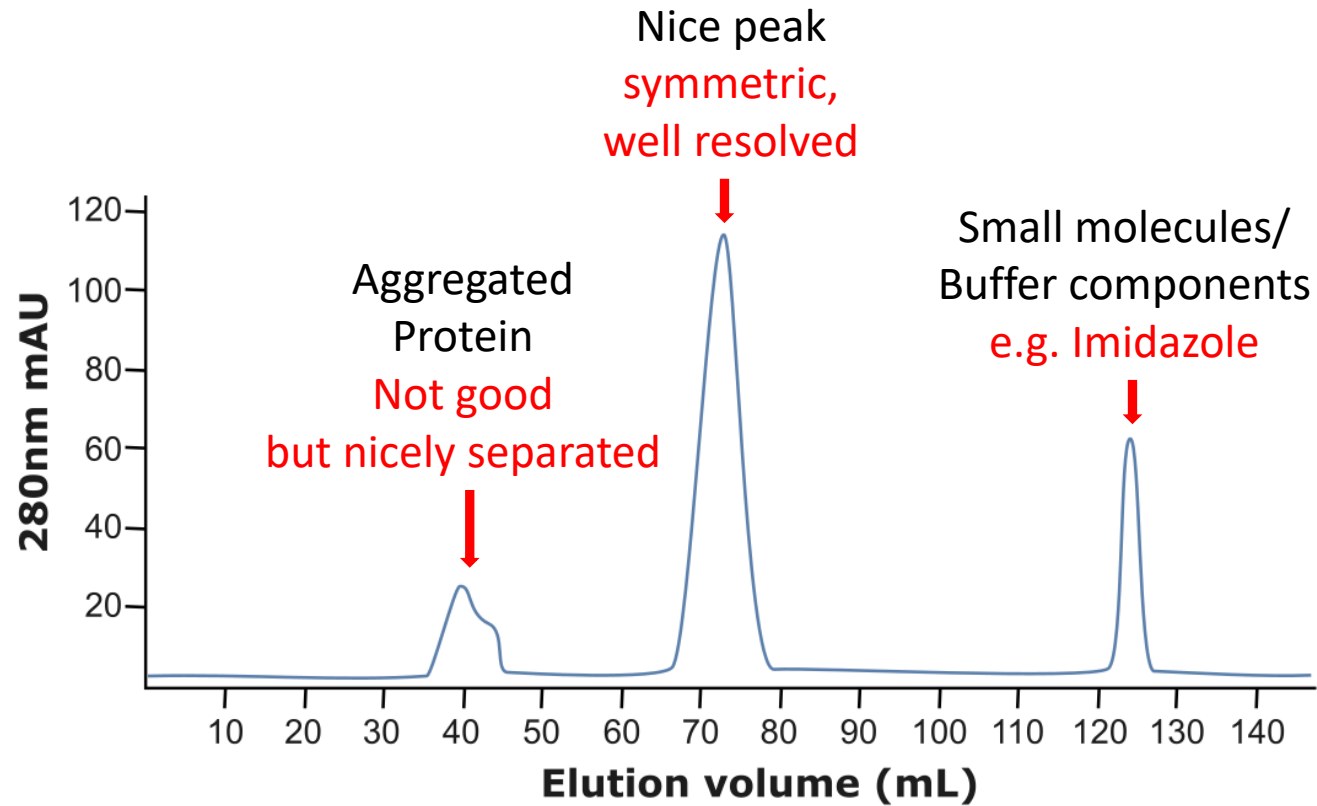
Interpreting Your SEC Elution Profile



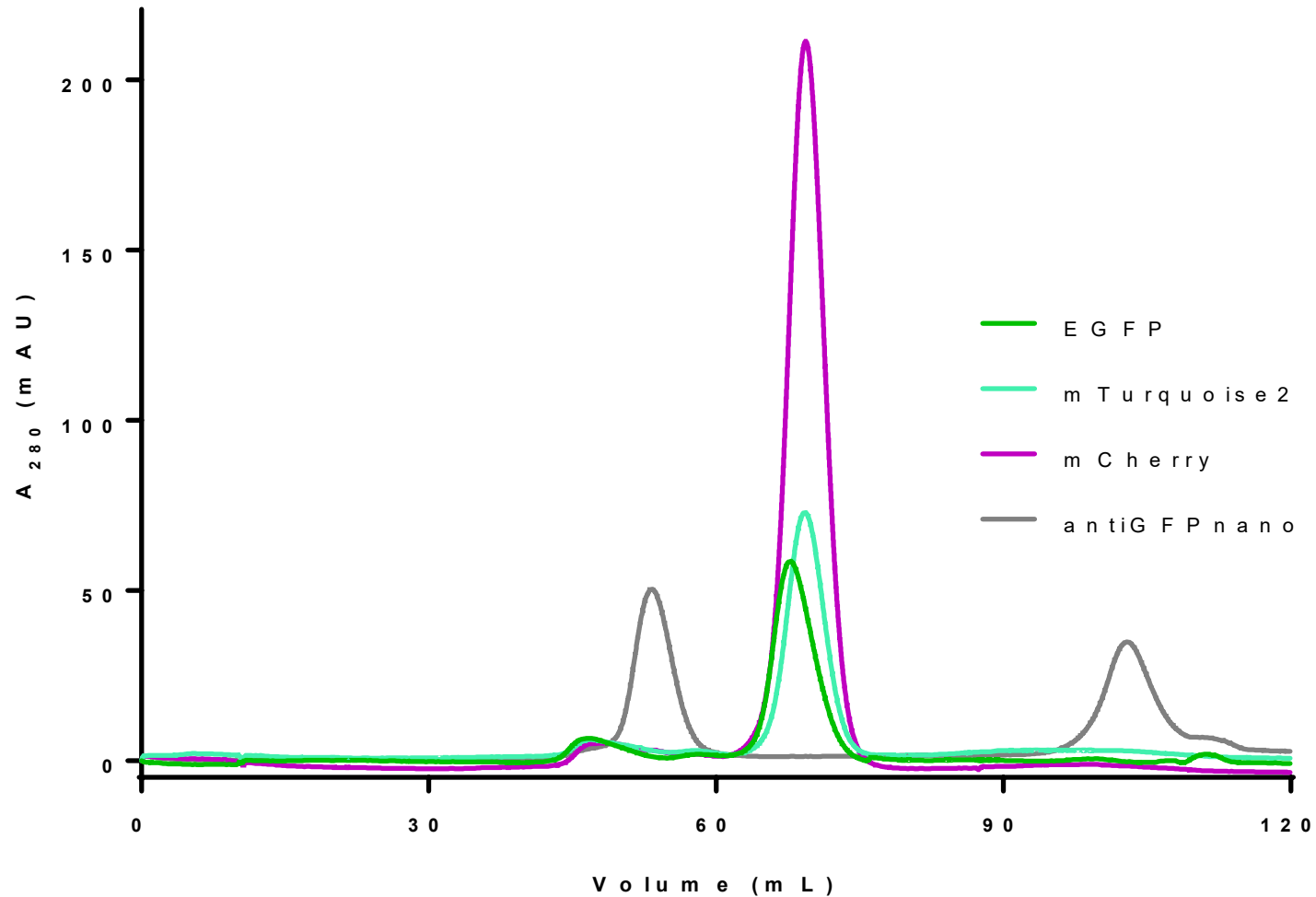
Interpreting Your SEC Elution Profile



Interpreting Your SEC Elution Profile



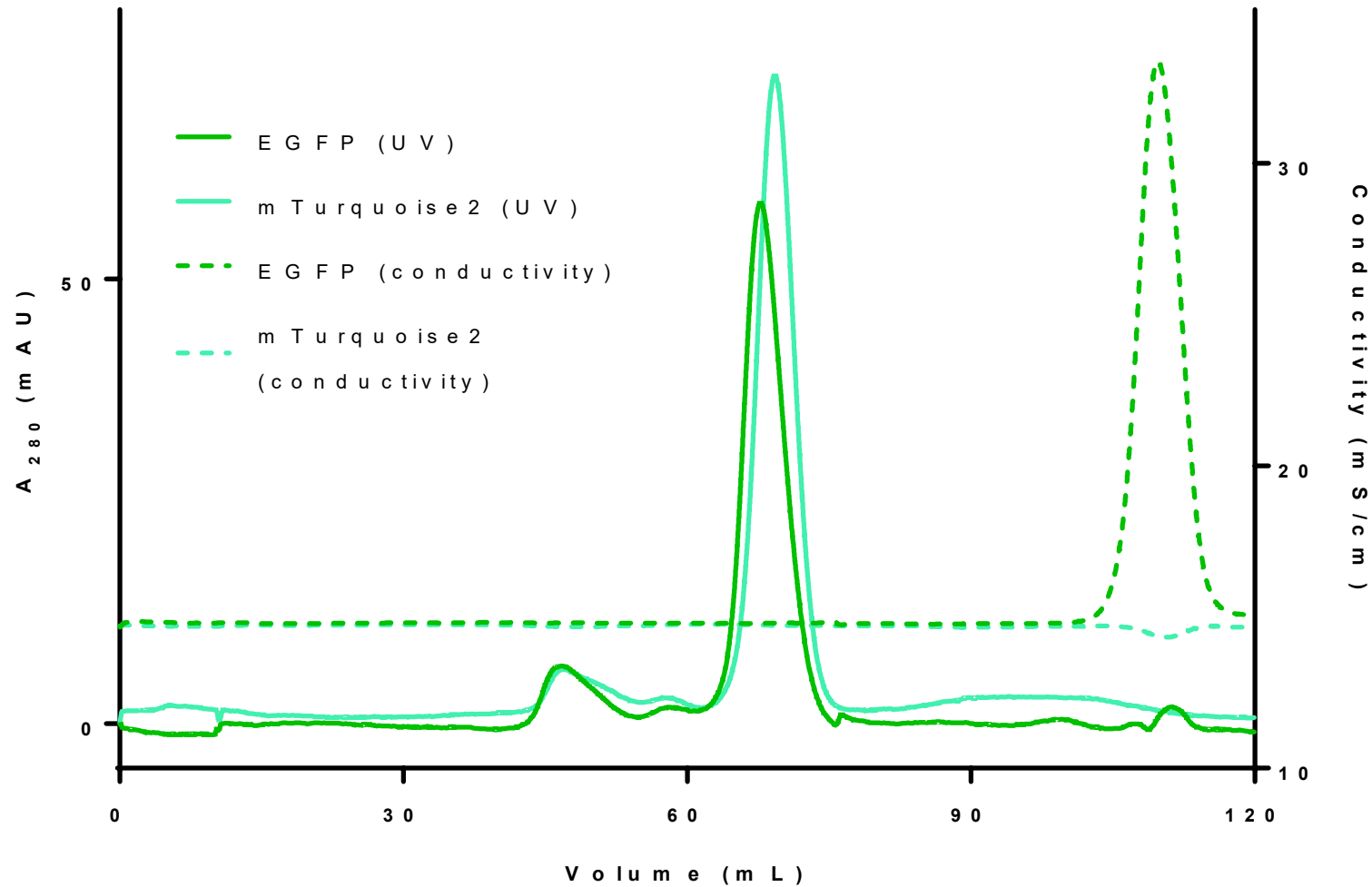
Your size-exclusion chromatography results



You can
download the
data from the
course website



Your size-exclusion chromatography results

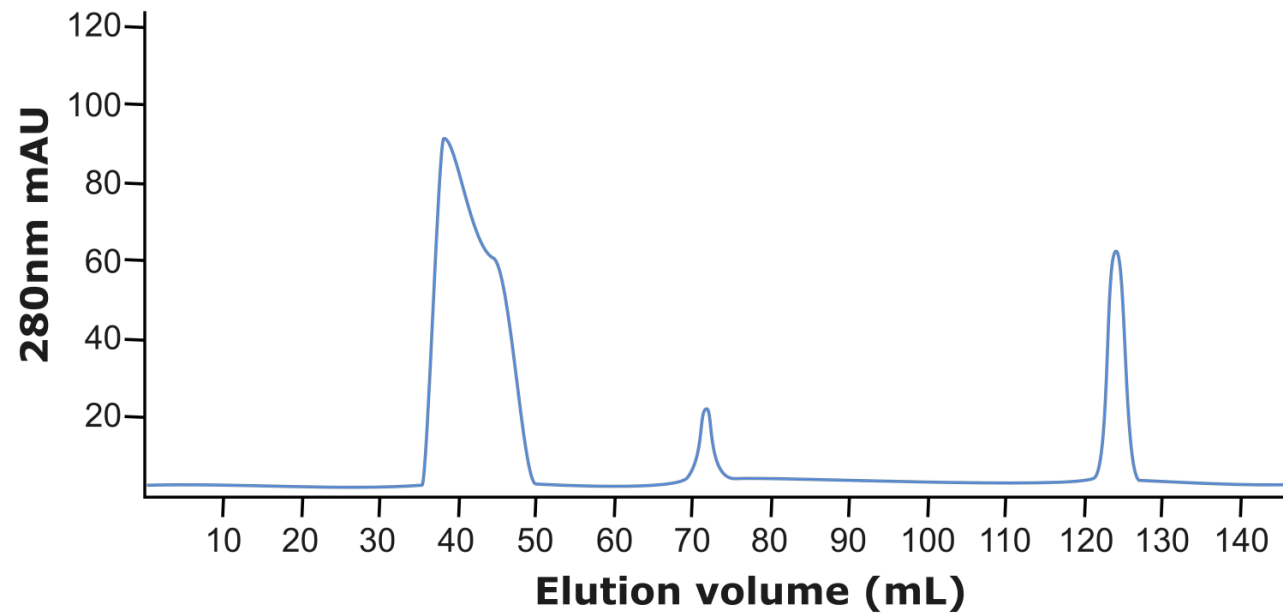


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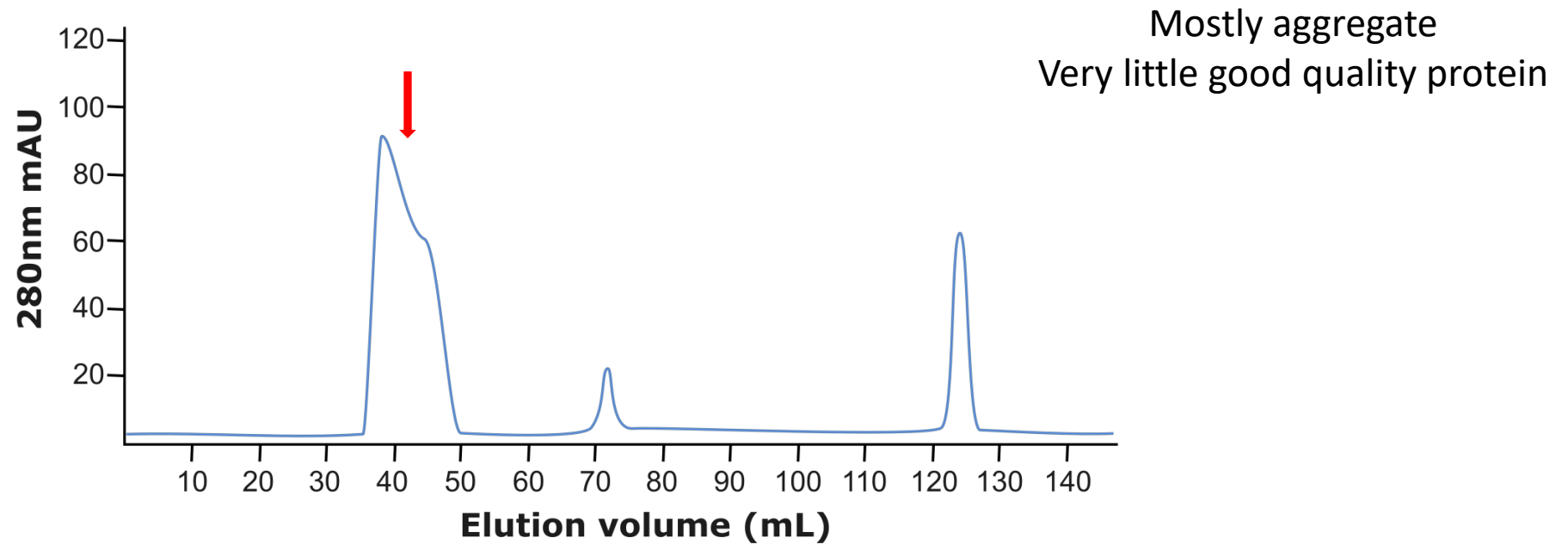
Interpreting SEC Elution Profiles

- What's going on here?



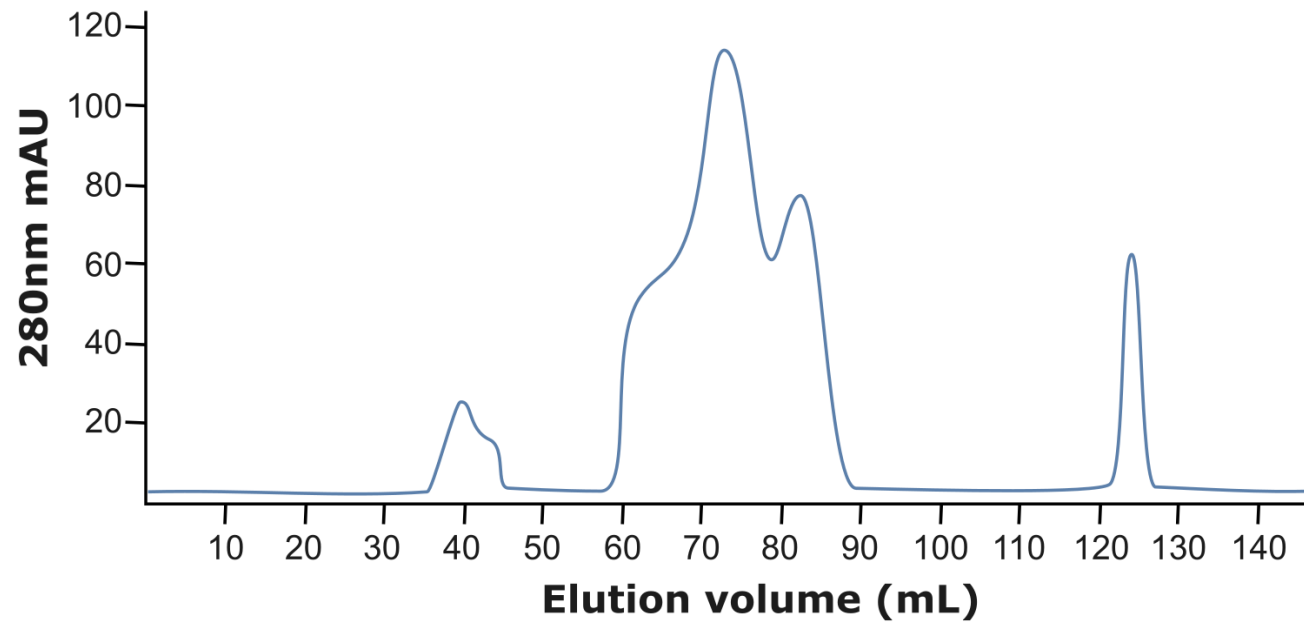
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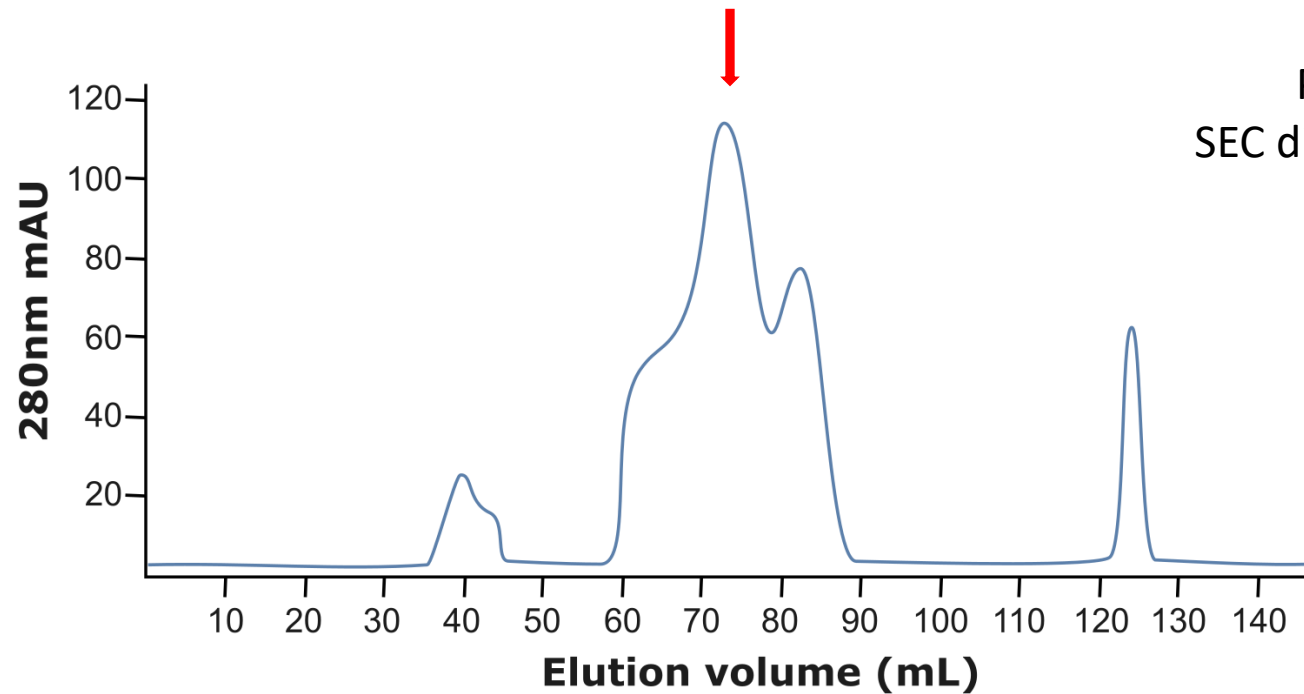
Interpreting SEC Elution Profiles

- What's going on here?



Interpreting SEC Elution Profiles

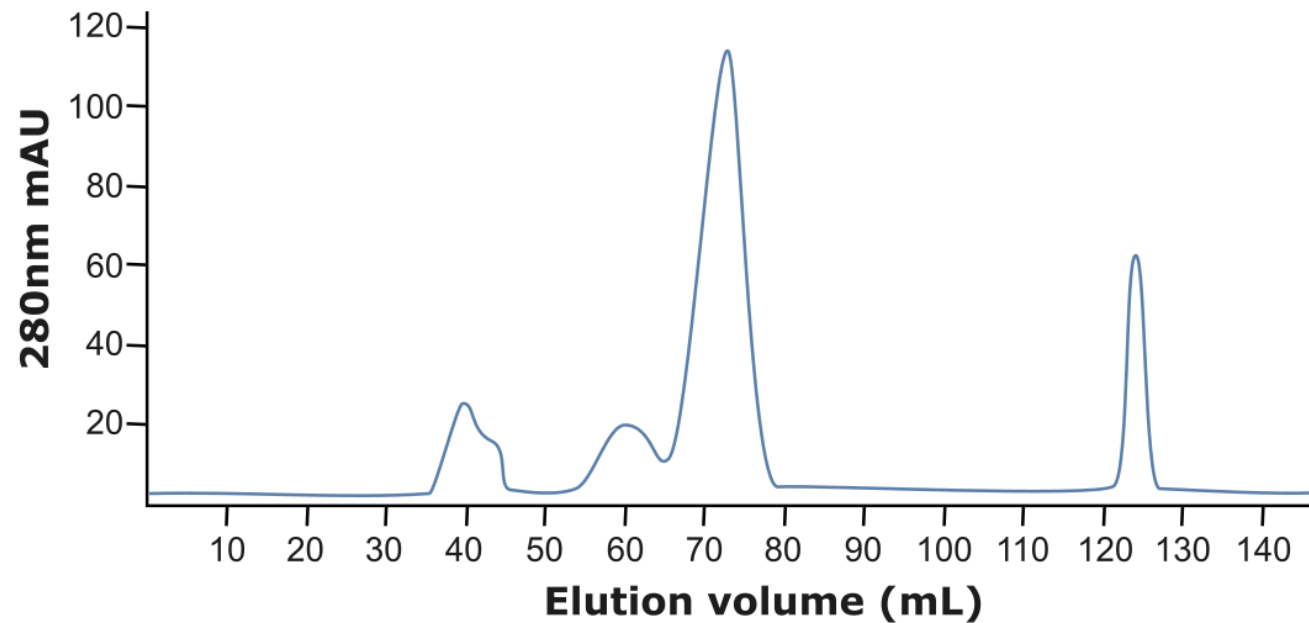
- What's going on here?



Protein not well resolved
SEC didn't really help purify protein

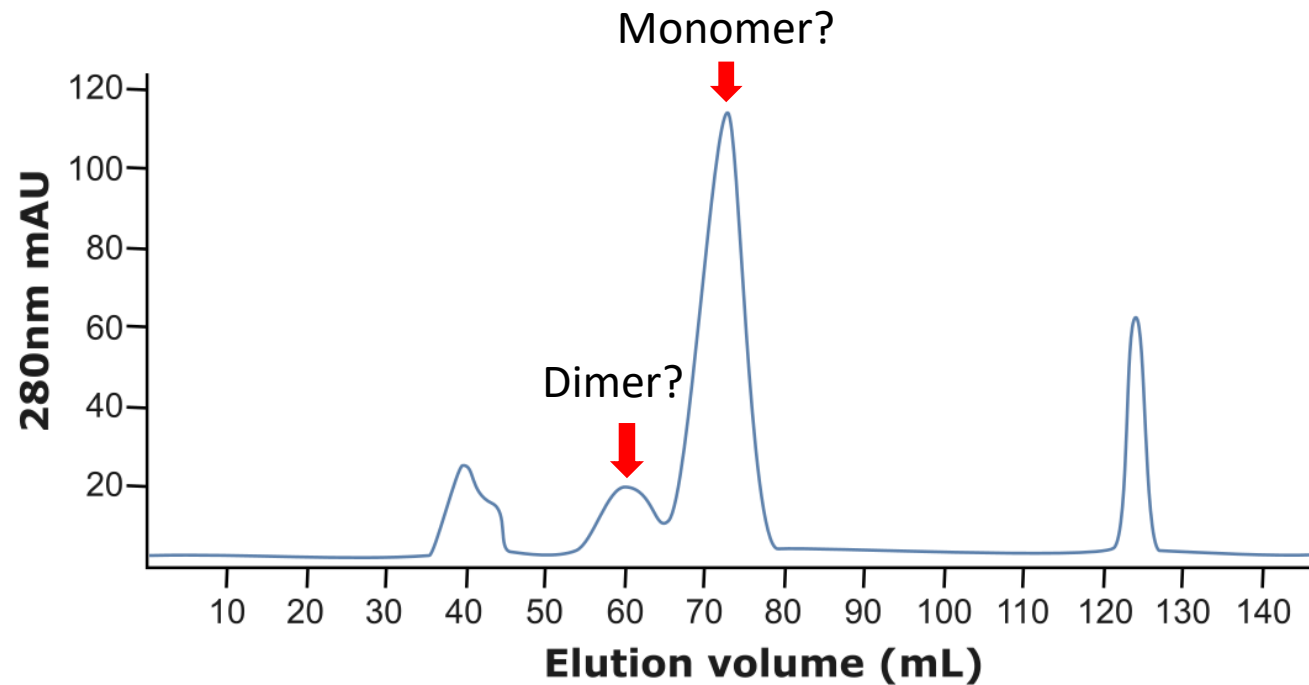
Interpreting SEC Elution Profiles

- What's going on here?



Interpreting SEC Elution Profiles

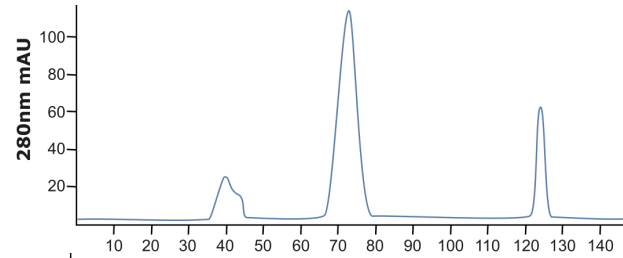
- What's going on here?



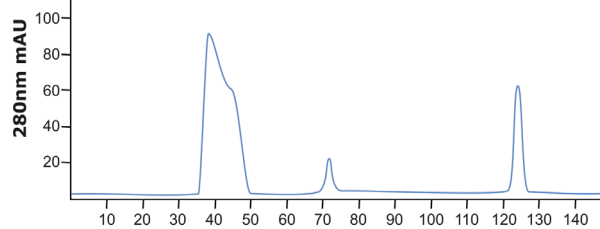
Could be forming
dimer and monomer



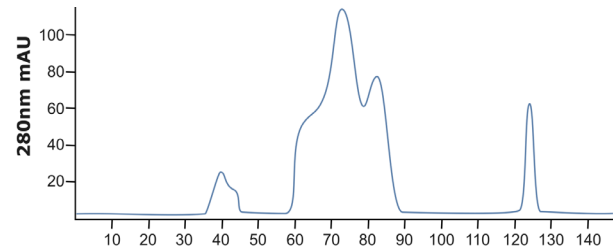
How might we check these hypotheses?



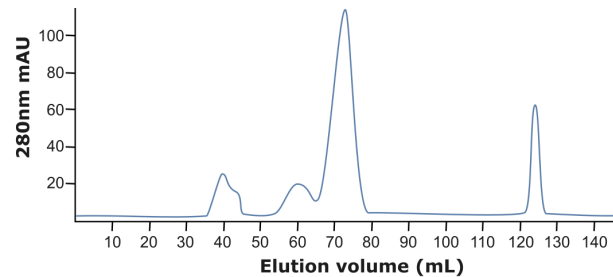
Nice, pure protein?



Aggregated protein?



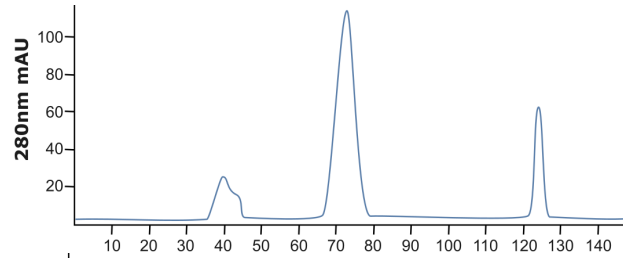
Impure protein?



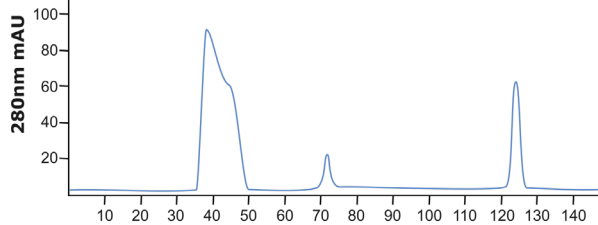
Dimer?



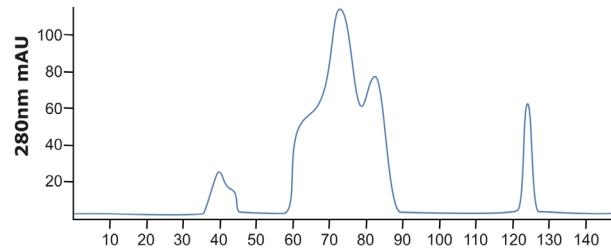
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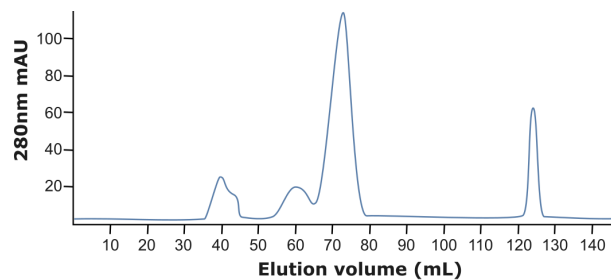
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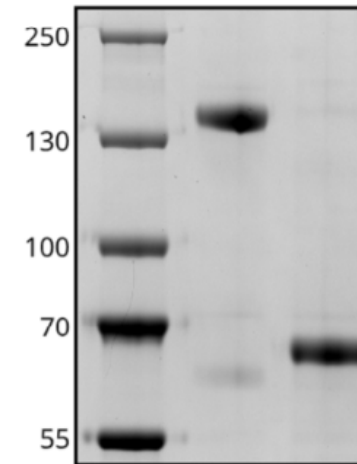


Impure protein?



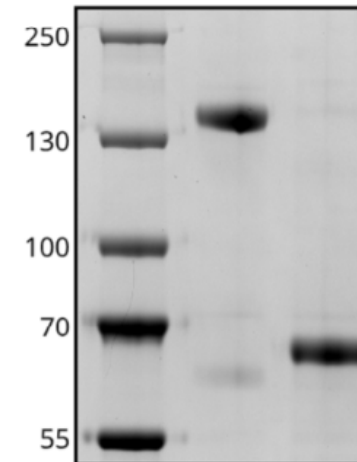
Dimer?

SDS-PAGE



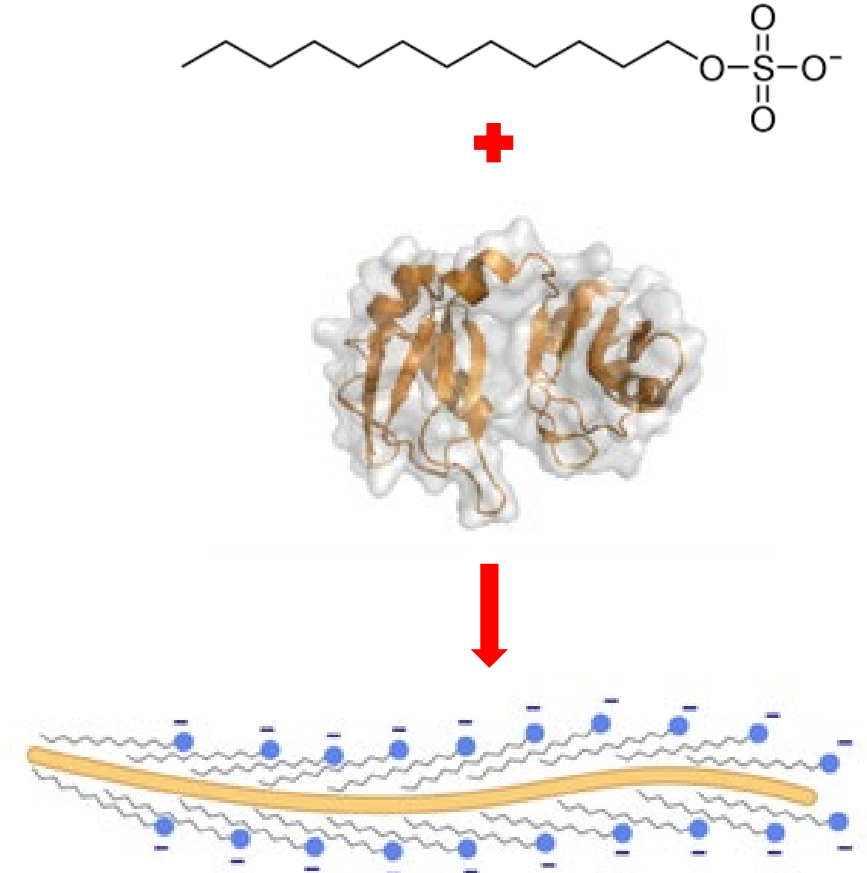
SDS-PAGE

- Allows separation of proteins based on MW
- Sample preparation
 - Why do we use SDS?
 - Why do we add DTT?
 - Why do we boil?
- Sample separation
 - Why do we use acrylamide?
 - How do we select the %acrylamide?
 - Why do we use a resolving gel?
 - Why use gradient or Tricine gels?



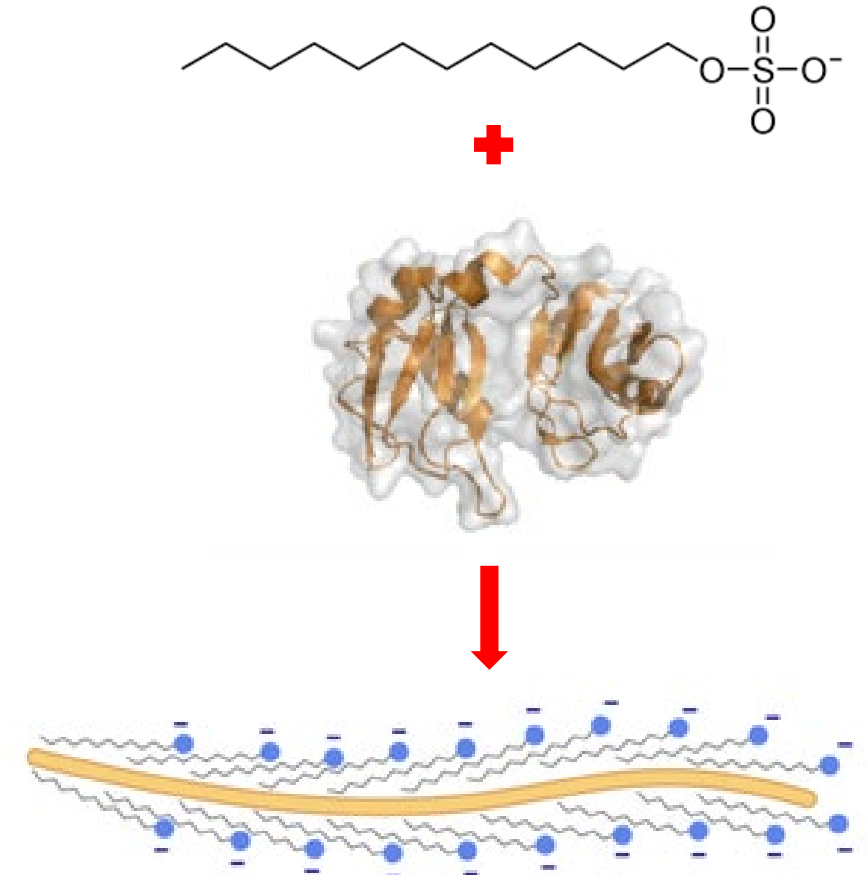
SDS – sodium dodecyl sulfate

- SDS possesses –ve charge
- About 1 SDS molecule per 2 amino acids
- Masks the protein's intrinsic charge
- Protein is now charged based on mass



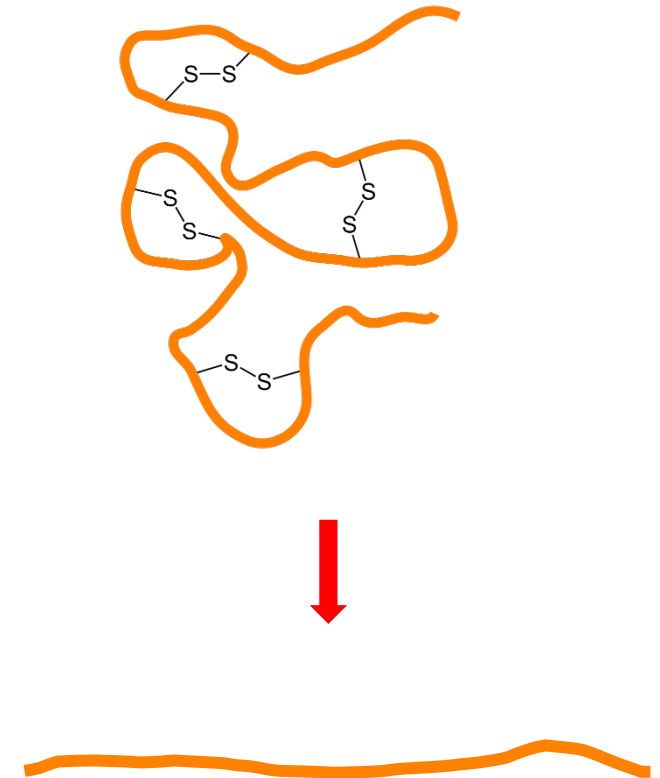
SDS – sodium dodecyl sulfate

- SDS possesses –ve charge
- About 1 SDS molecule per 2 amino acids
- Masks the protein's intrinsic charge
- Protein is now charged based on mass
- Amphipathic nature of SDS allows it to unfold polar and non-polar structures
- Highly denaturing, disrupt complexes as well as protein fold



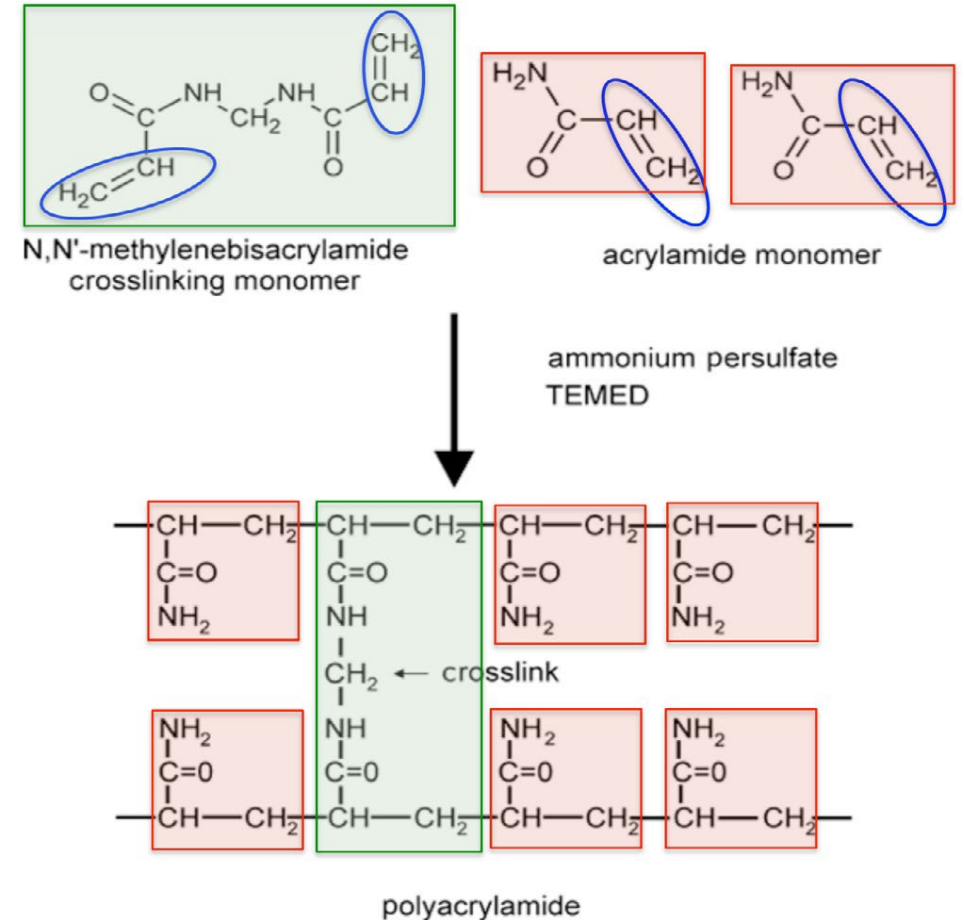
Why add DTT and why boil?

- DTT
 - Reduce disulfide bonds
 - If these are retained, the sample will not run according to its mass
- Boiling
 - Remove any SDS-resistant structure
 - Ensure sample is fully unfolded
- The proteins will now run according to MW



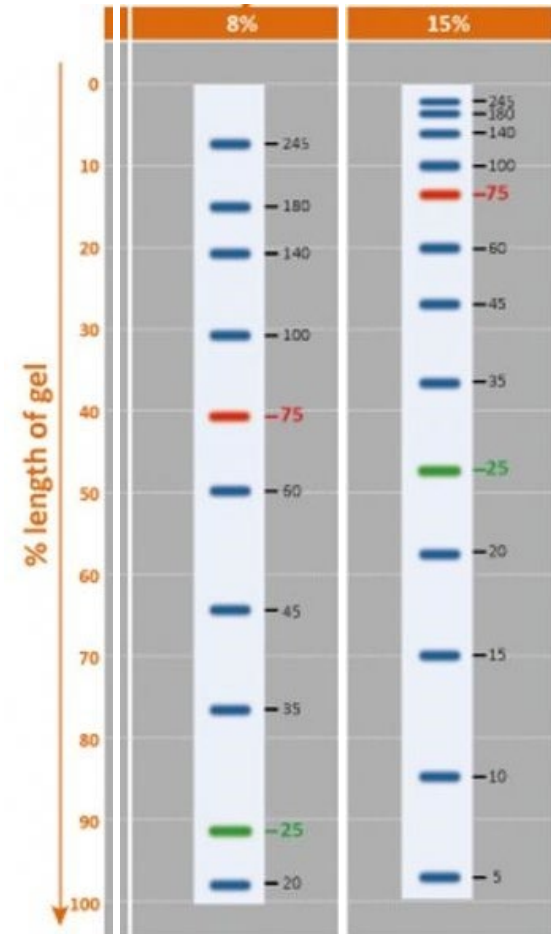
Acrylamide to polyacrylamide

- Acrylamide/Bis solution
- Free radical polymerisation
- Polymerisation of bis/acrylamide is carried out by adding:
 - *catalyst* TEMED and
 - *radical initiator* ammonium persulfate (APS)



Protein separation

- Negatively charged molecules migrate toward the +ve anode
- The polymerised acrylamide acts like a sieve
- The % acrylamide determines what sizes are best resolved

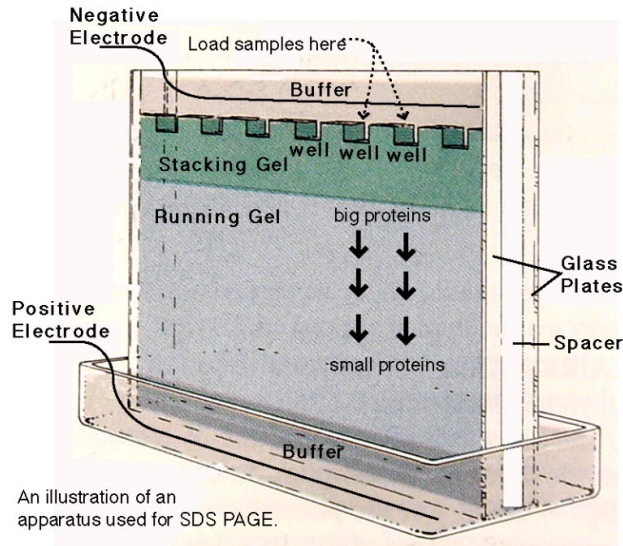


Protein separation

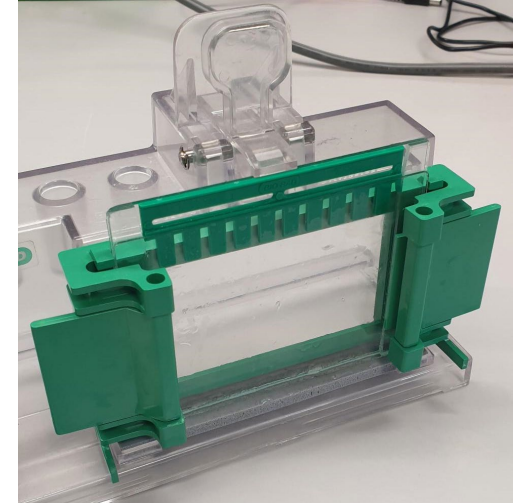
- Negatively charged molecules migrate toward the +ve anode
- The polymerised acrylamide acts like a sieve
- The % acrylamide determines what sizes are best resolved
- Higher % better for small proteins

Acrylamide (%)	Mw Range (KDa)
7	50–500
10	20–300
12	10–200
15	3–100

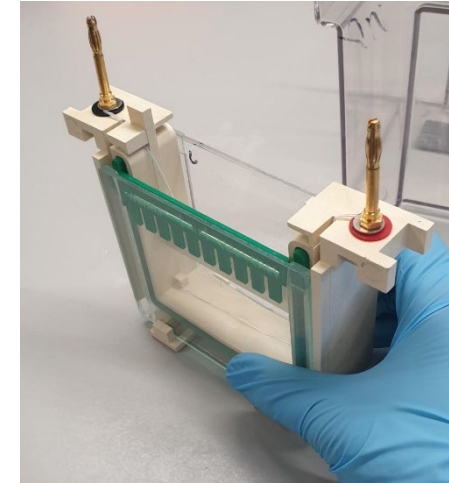
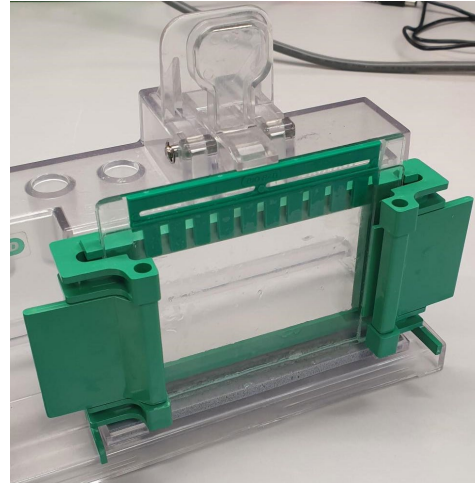
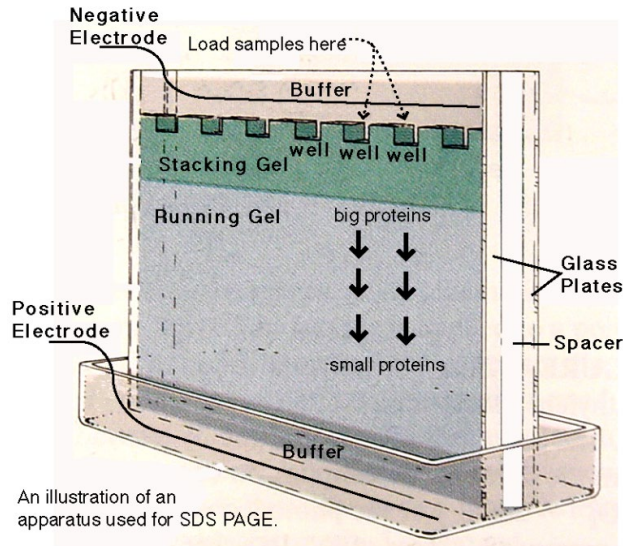
Assembling and pouring the gel



Take two glass plates
Use a holder to clamp them
Use a stand to seal the bottom
Pour in the resolving gel mix
Let it set
Pour in the stacking gel mix with a well comb



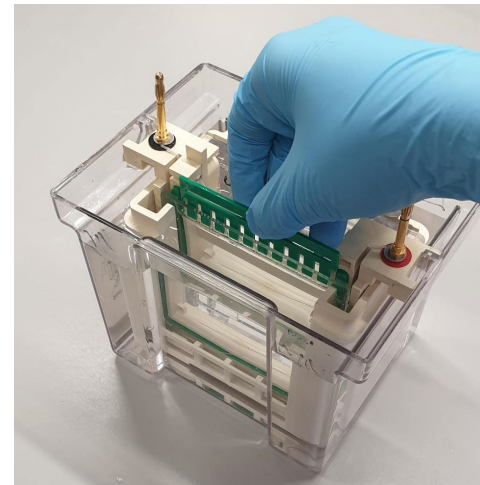
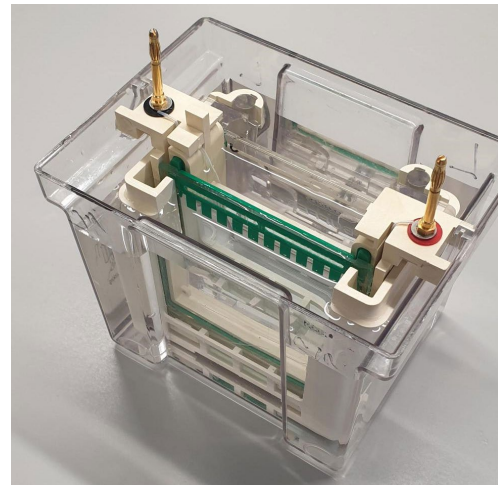
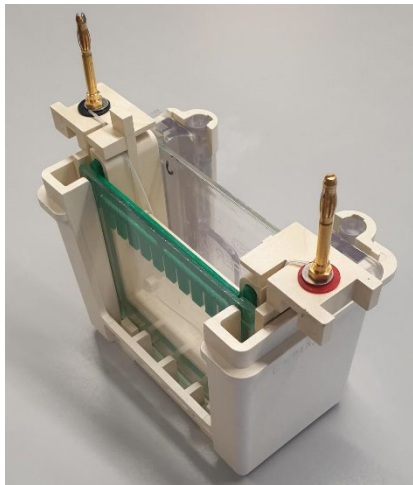
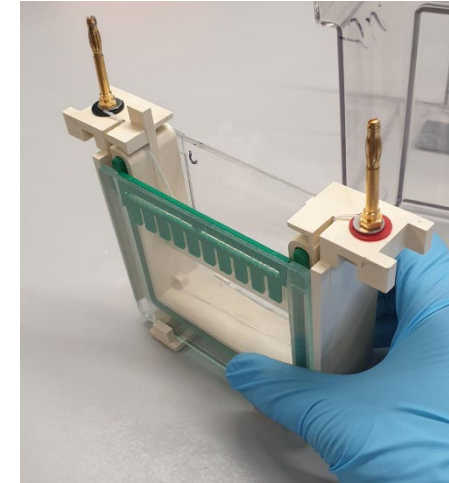
Assembling and pouring the gel



Once it's set, remove it from the stand and holder
Insert it into the central tank apparatus
Add buffer dam plate to seal the chamber

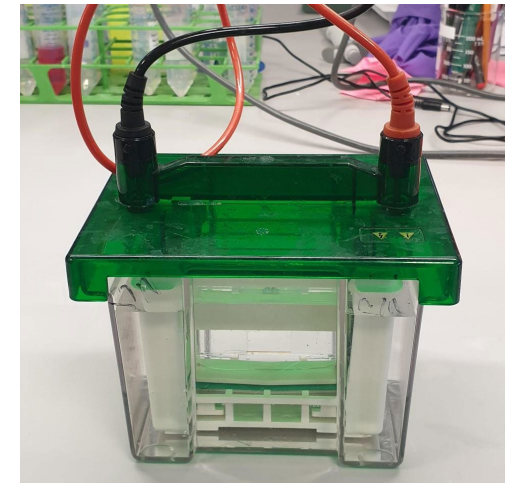
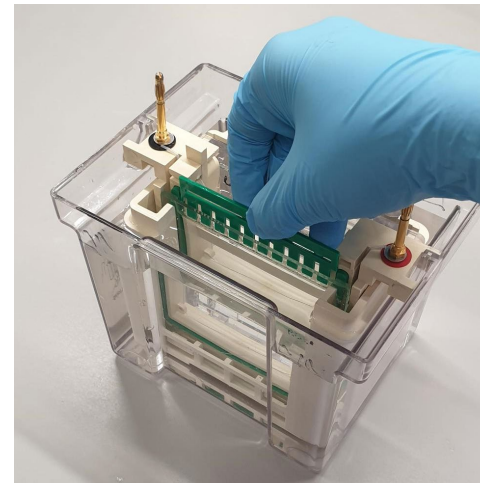
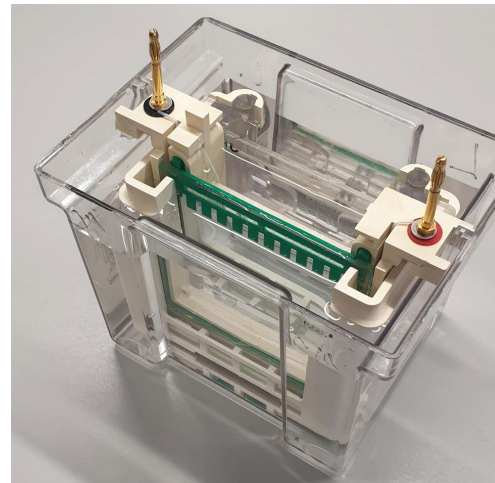
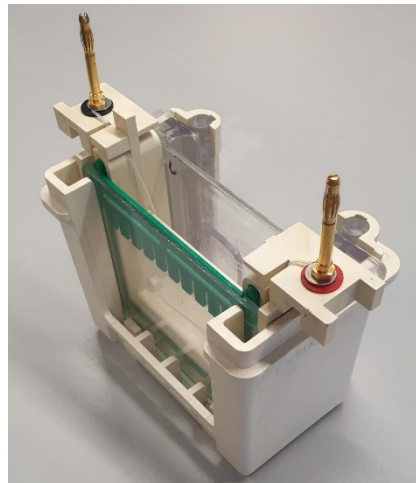
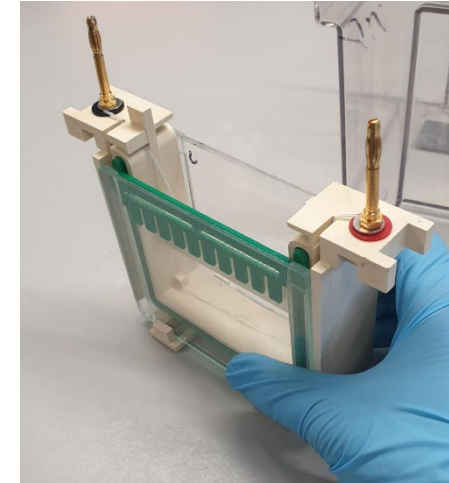
Assembling and pouring the gel

Assemble the remainder of the gel tank
Add running buffer to the tank
Remove the well comb



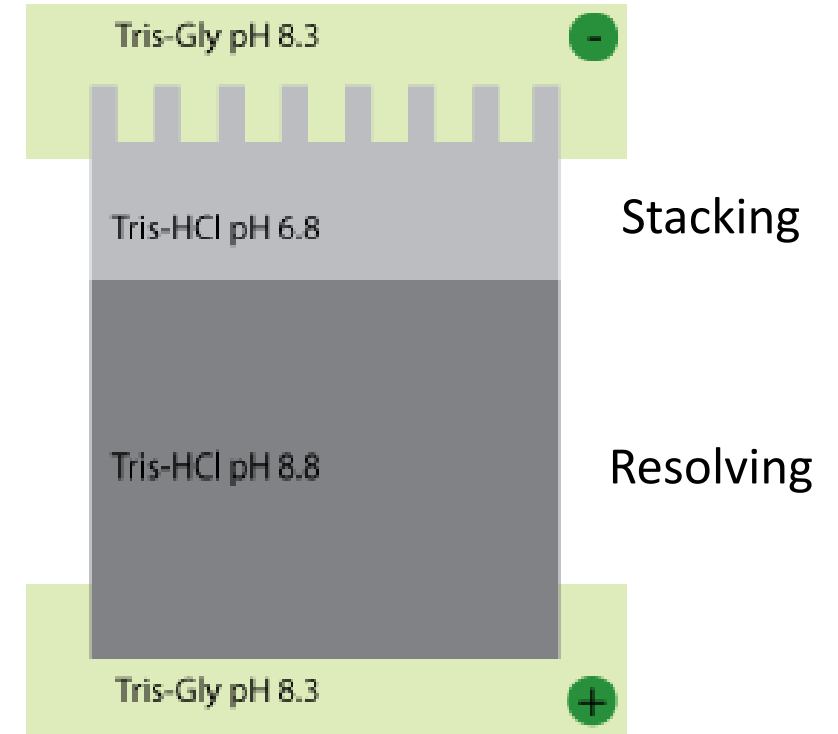
Assembling and pouring the gel

- Assemble the remainder of the gel tank
- Add running buffer to the tank
- Remove the well comb
- Add samples
- Run the gel



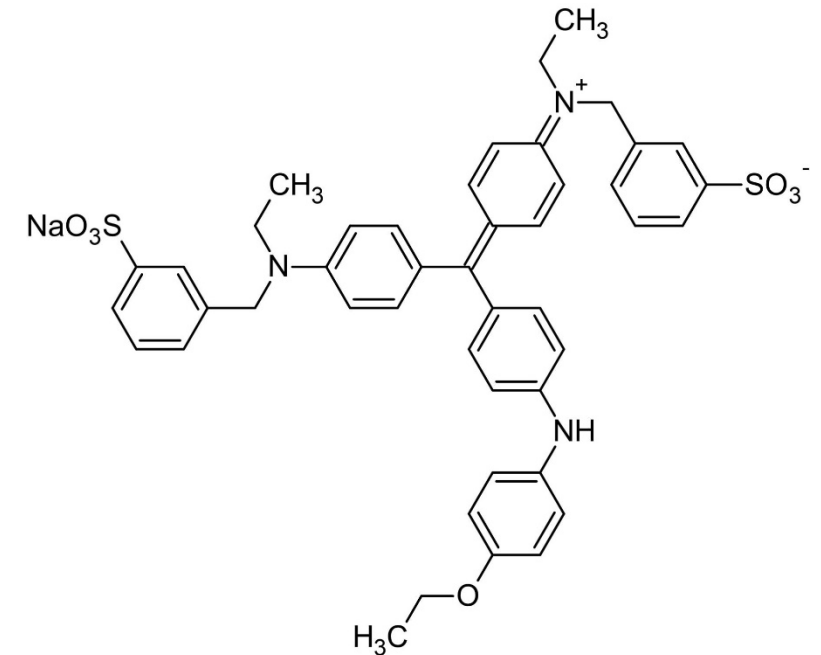
Protein separation – stacking and resolving gels

- “Discontinuous Buffer System”
 - Buffer in the tank and gel differ in terms of composition and pH
- Stacking gel: pH 6.8 and low % acrylamide
 - Don't want proteins to separate (low %)
 - Glycine –ve at pH 8.3, enter low pH stacking gel, switch to neutral so moves very slowly
 - The Cl⁻ ions from Tris-HCl highly mobile
 - Proteins intermediate mobility, concentrated into narrow band between these two fronts

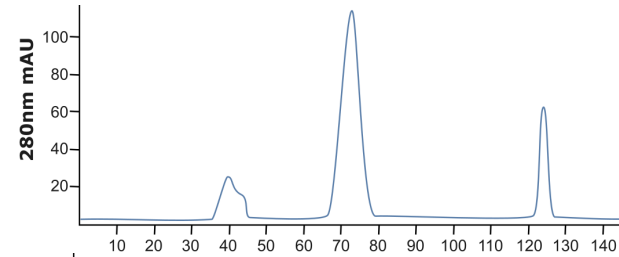


Gel staining

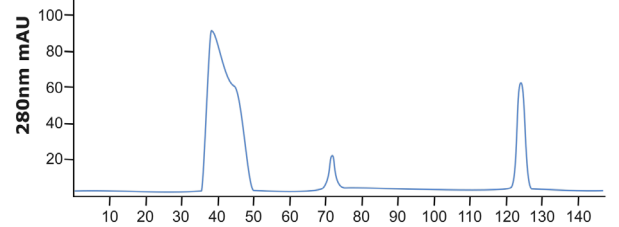
- Coomassie
 - Most common
 - Cheap
 - Ionic interaction between sulfonic acid groups and positive protein amine groups
- Silver stain or Sypro Orange stain
 - More sensitive
- Unlike Western blots, you see everything



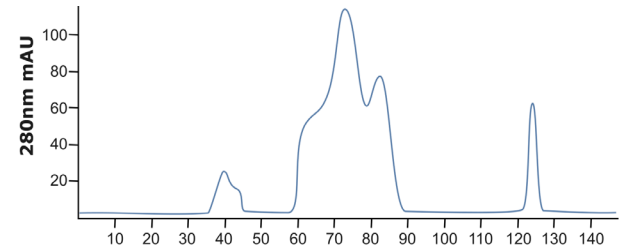
How can we use SDS-PAGE to check our SEC samples?



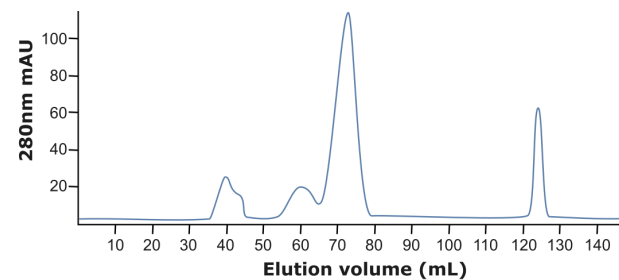
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Aggregated protein?



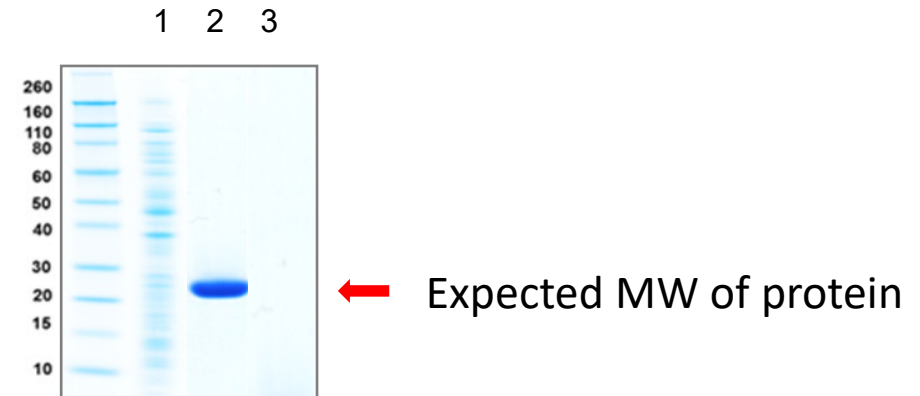
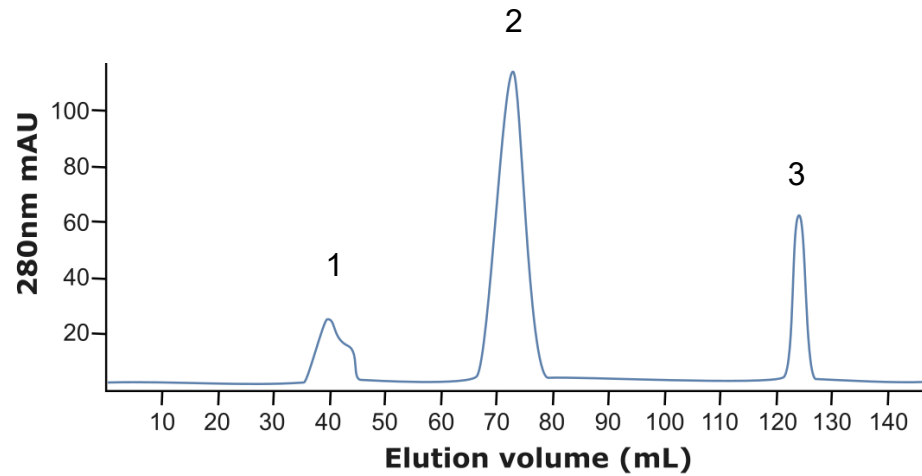
Impure protein?



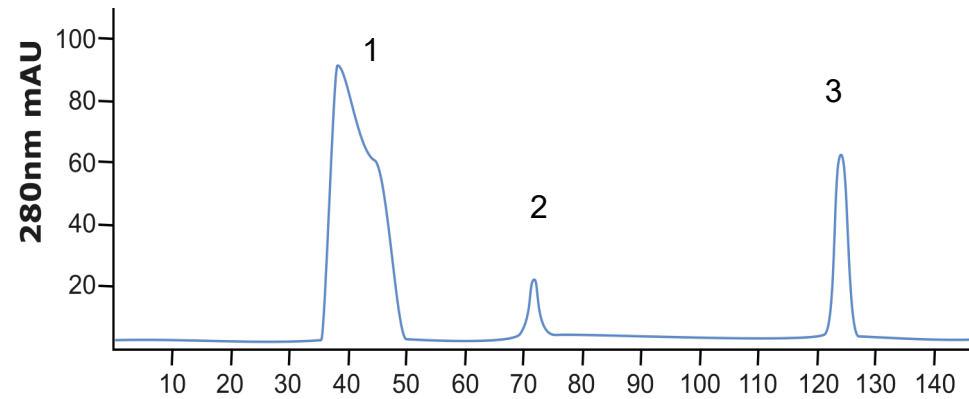
Dimer?



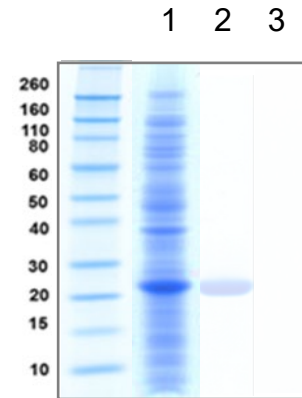
SDS-PAGE Analysis



SDS-PAGE Analysis

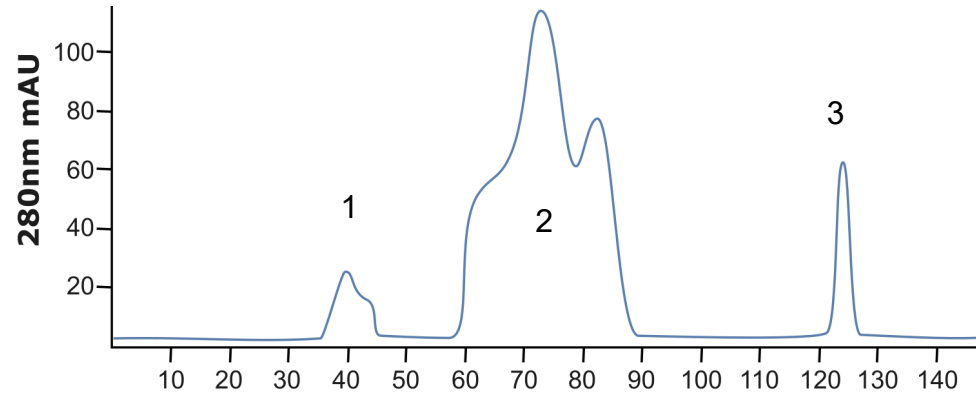


Aggregated protein?

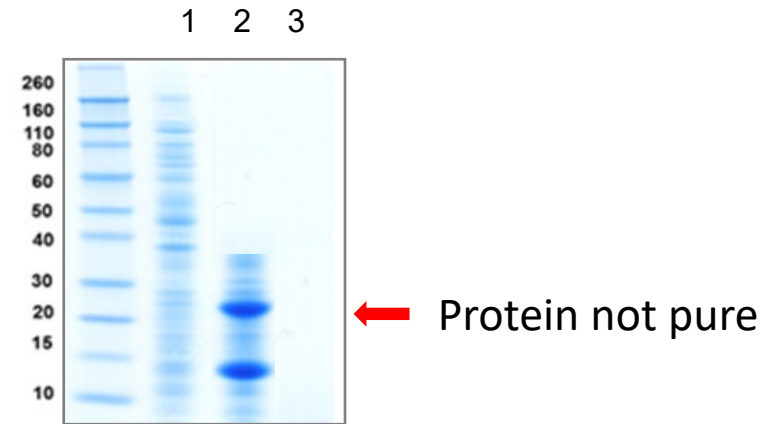


← Most of protein in aggregate peak

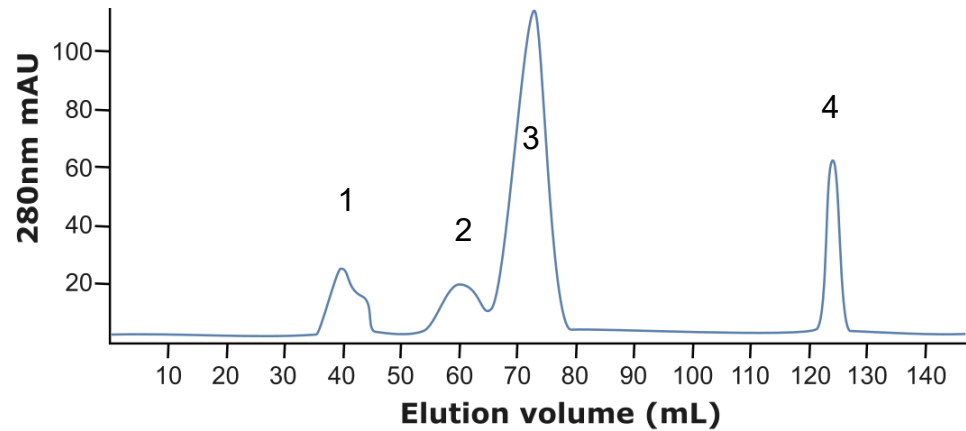
SDS-PAGE Analysis



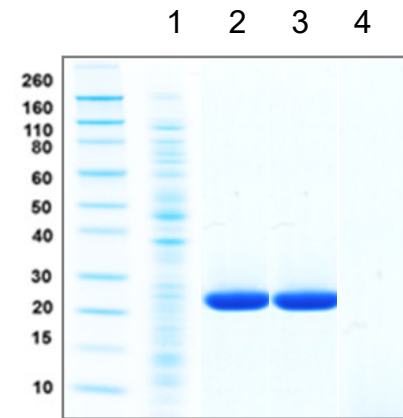
Impure protein?



SDS-PAGE Analysis

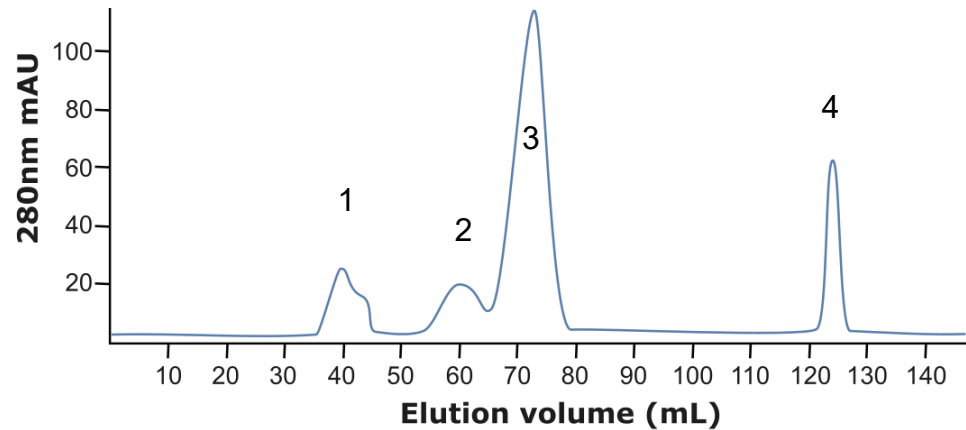


Dimer?

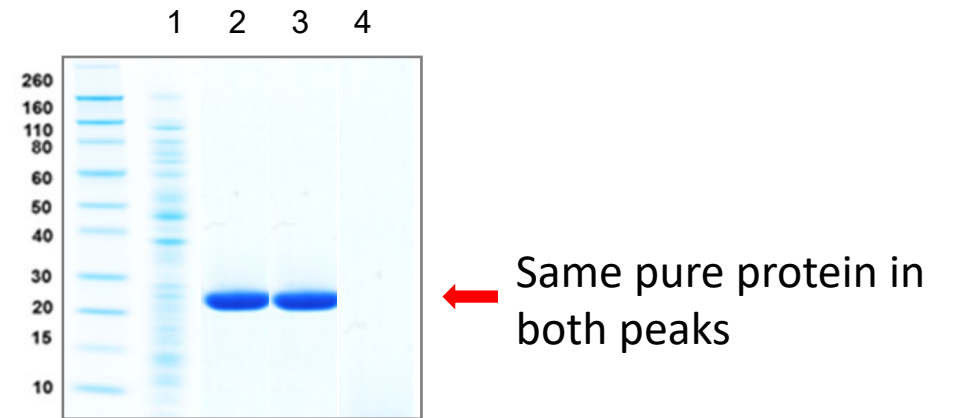


← Same pure protein in both peaks

SDS-PAGE Analysis



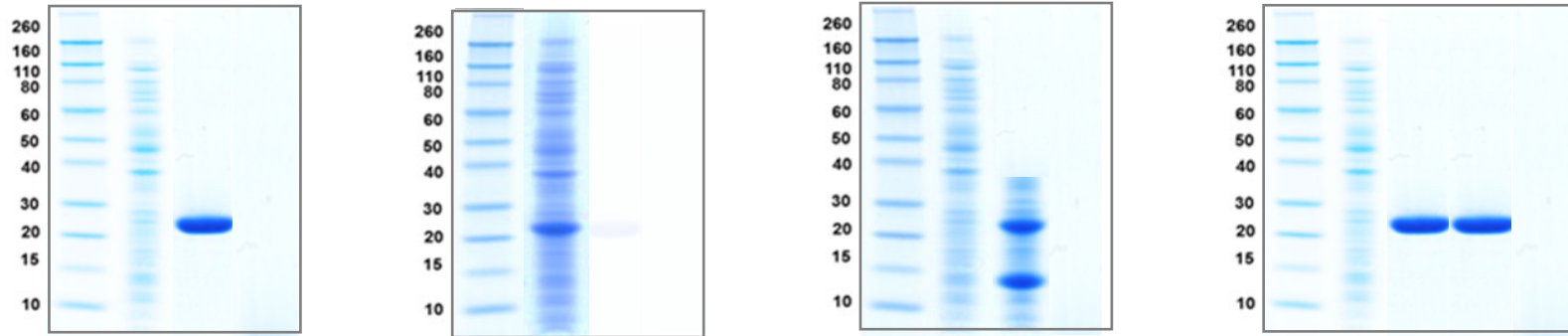
Dimer?



Remember SDS-PAGE denatures proteins so dimeric protein will run as a monomer

SDS-PAGE Analysis

- Combined with SEC, normal SDS-PAGE can tell us a lot



- There are some other useful things we can learn using SDS-PAGE

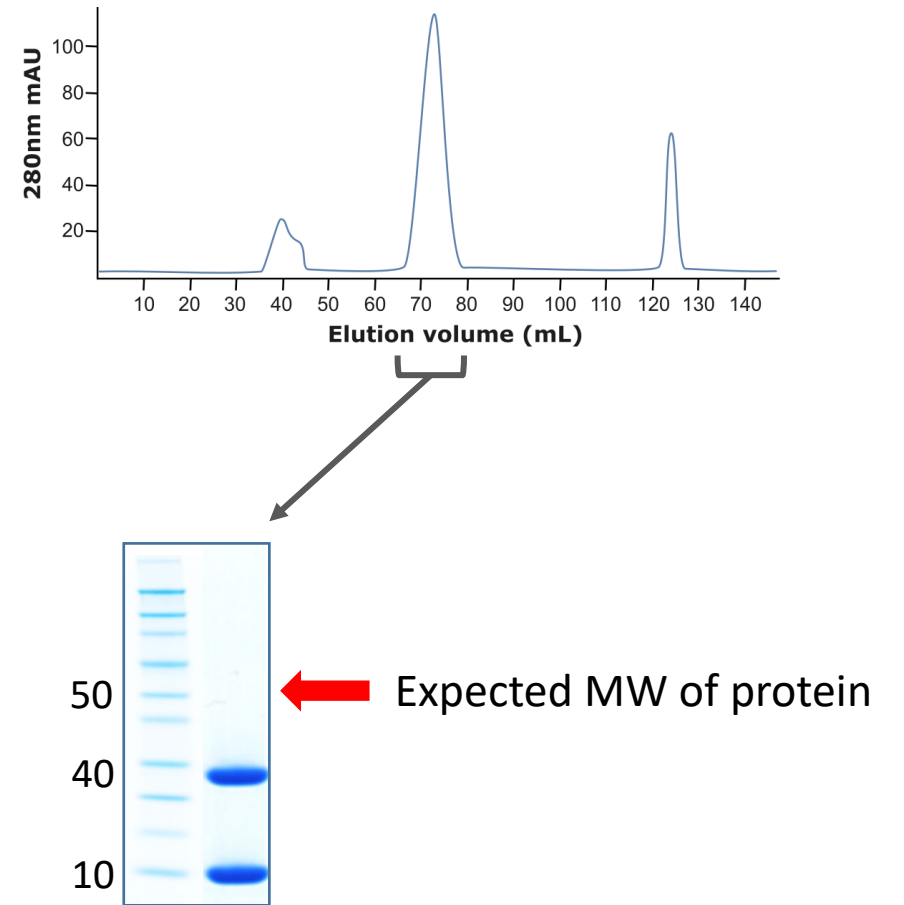
SDS-PAGE Analysis

- Based on our knowledge of how SDS-PAGE works we can answer additional questions:
 - Has my protein been clipped/processed?
 - Does my protein have disulfide bonds?
 - Does my protein have any post-translational modifications such as glycosylation or phosphorylation?



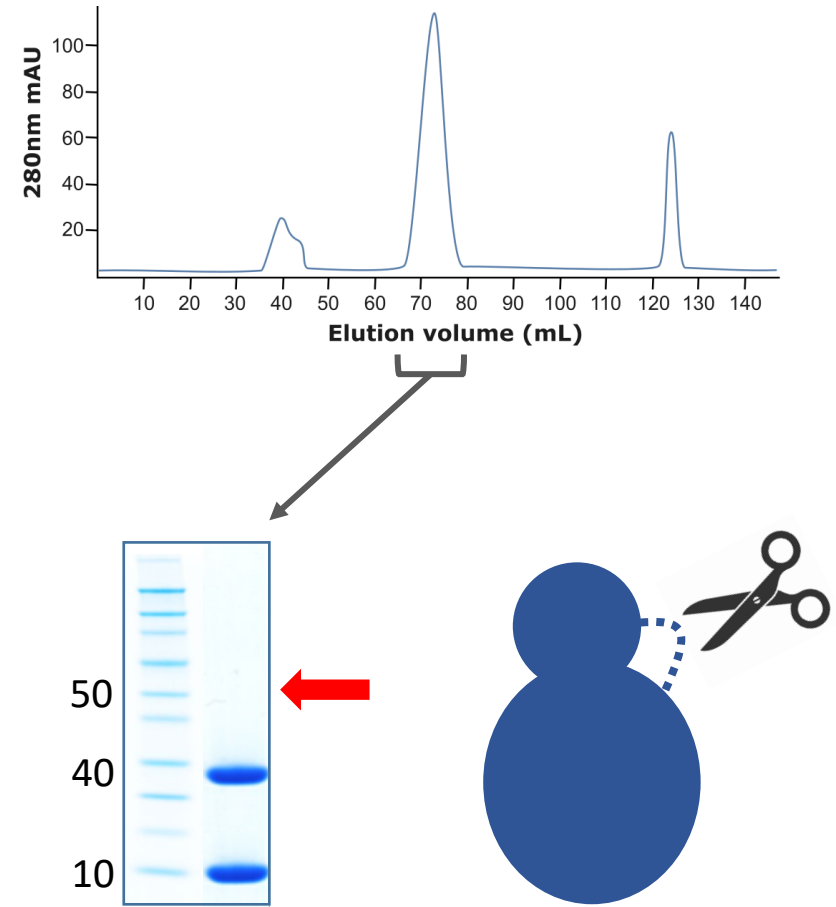
Protein processing – clipped loops

- What if I had a lovely peak off SEC but it's two bands of the wrong size



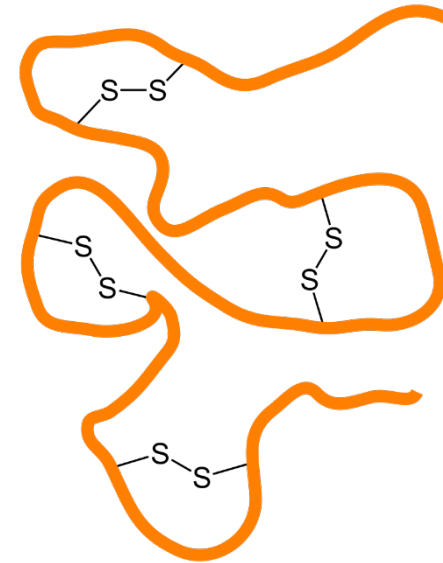
Protein processing – clipped loops

- What if I had a lovely peak off SEC but it's two bands of the wrong size
- SDS-PAGE unfolds and separates protein so if there is a clipped surface loop you're protein can be correctly folded but look like this



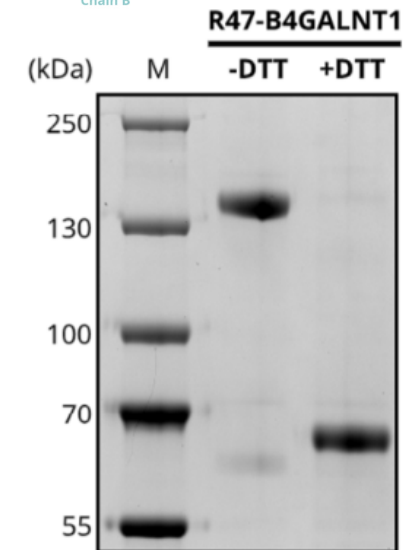
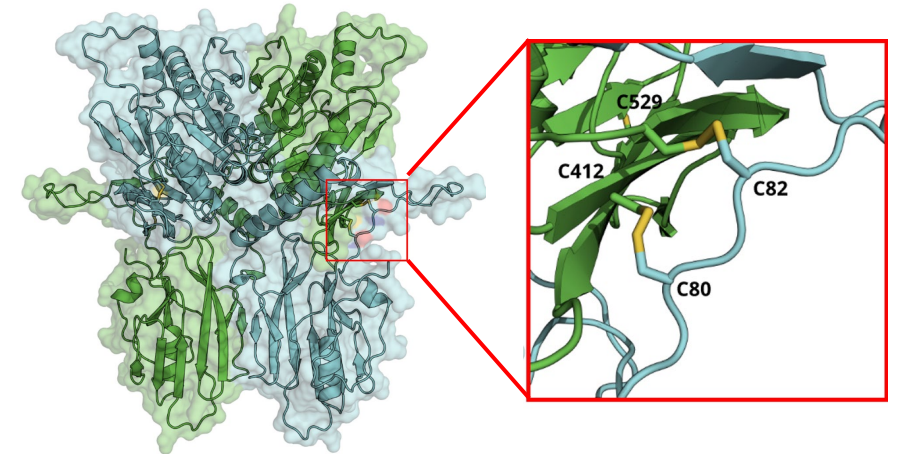
Disulfide bond formation

- When preparing SDS-PAGE samples we not only boil in SDS, we also add reducing agent (DTT) to break disulfide bonds
- This makes sure they run at their correct molecular weight



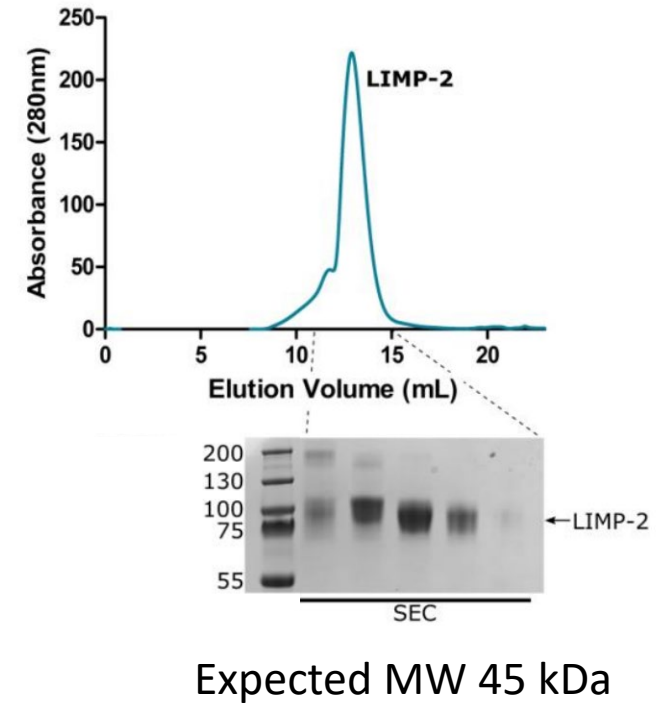
Disulfide bond formation

- Non-reducing SDS-PAGE (without DTT) can reveal if your protein forms:
 - Intramolecular disulfide bonds
 - Intermolecular disulfide bonds



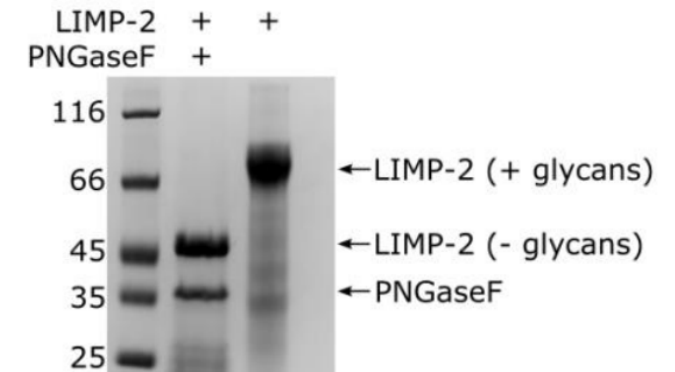
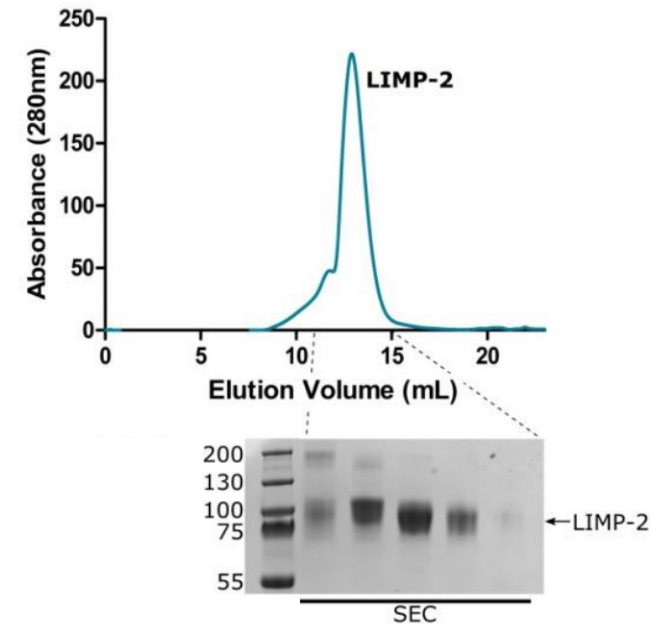
Post-translational Modifications

- If your protein is glycosylated on SDS-PAGE it might:
 - run much larger than expected
 - appear smeary



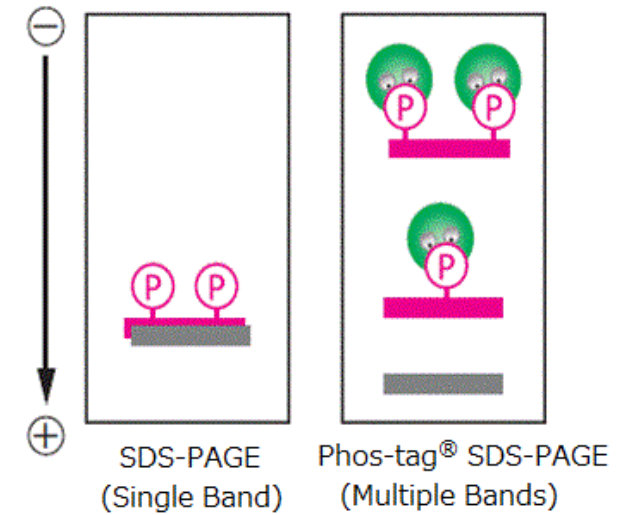
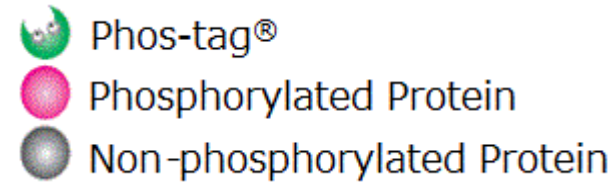
Post-translational Modifications

- If your protein is glycosylated on SDS-PAGE it might:
 - run much larger than expected
 - appear smeary
- Deglycosylation using PNGaseF can confirm this, reducing your protein to its predicted MW



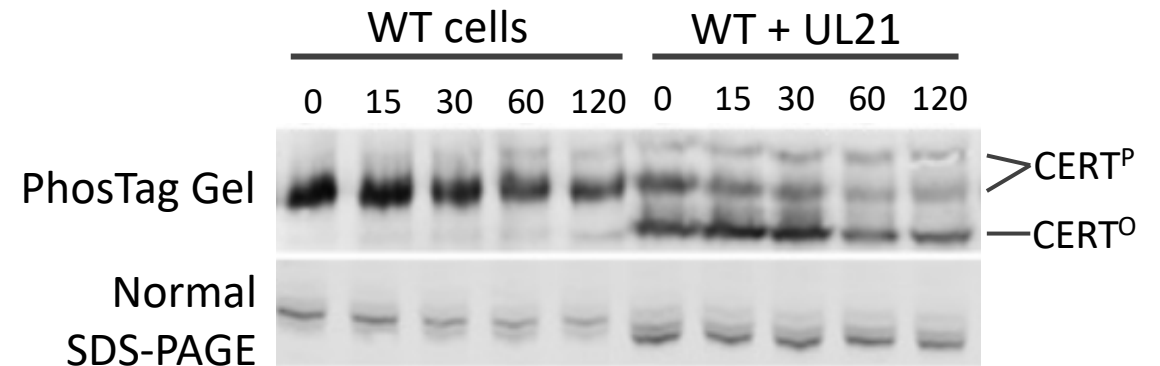
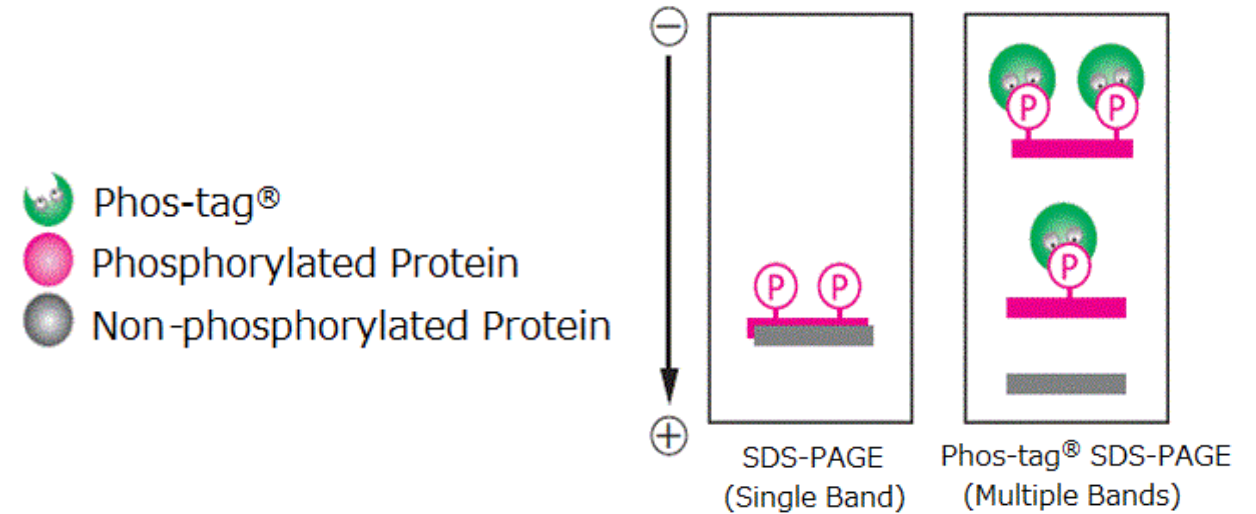
Post-translational Modifications - phosphorylation

- Add a Phostag reagent into SDS-PAGE gel mix
- Retards phosphorylated proteins so run higher



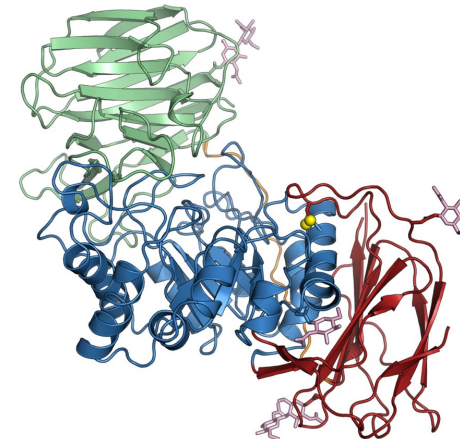
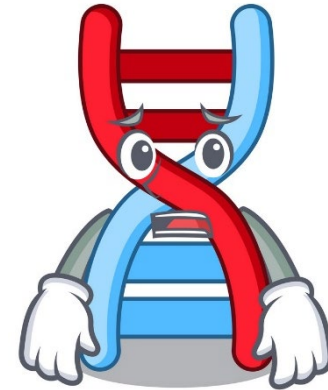
Phoshtag Gels

- Add a Phostag reagent into SDS-PAGE gel mix
- Retards phosphorylated proteins so run higher
- The more phospho sites the higher they run



What can't SDS-PAGE tell us about our sample?

- Using SDS-PAGE we can't:
 - Identify DNA contamination
 - Reveal if your protein is actually folded
- How do we check this?



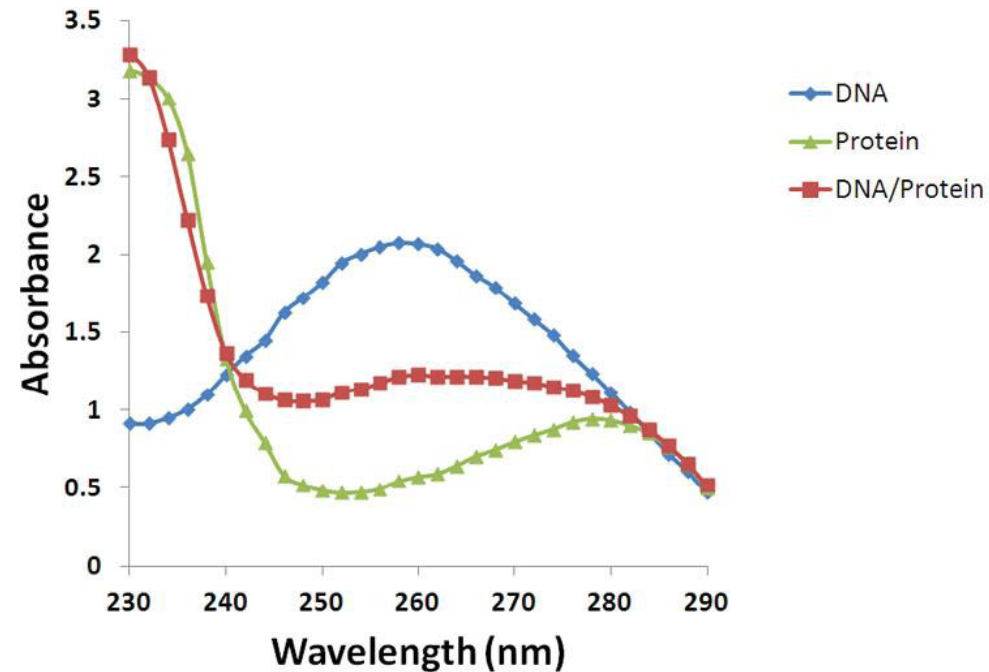
Estimating Purity Using UV Spectrophotometry

- UV spectrophotometry and the A260:280 ratio



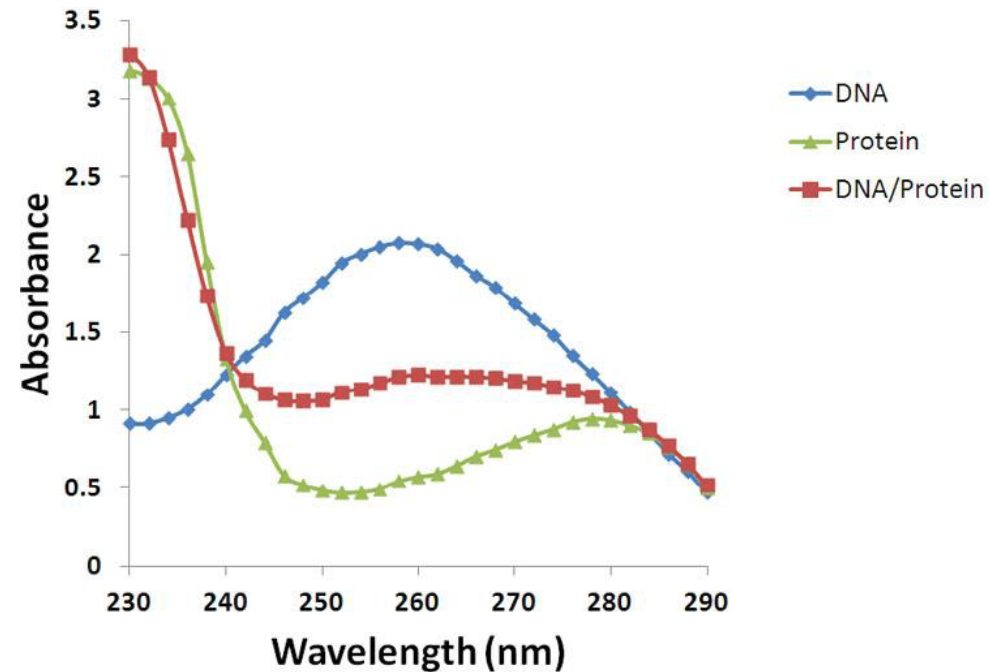
Estimating Purity Using UV Spectrophotometry

- UV spectrophotometry and the A260:280 ratio



Estimating Purity Using UV Spectrophotometry

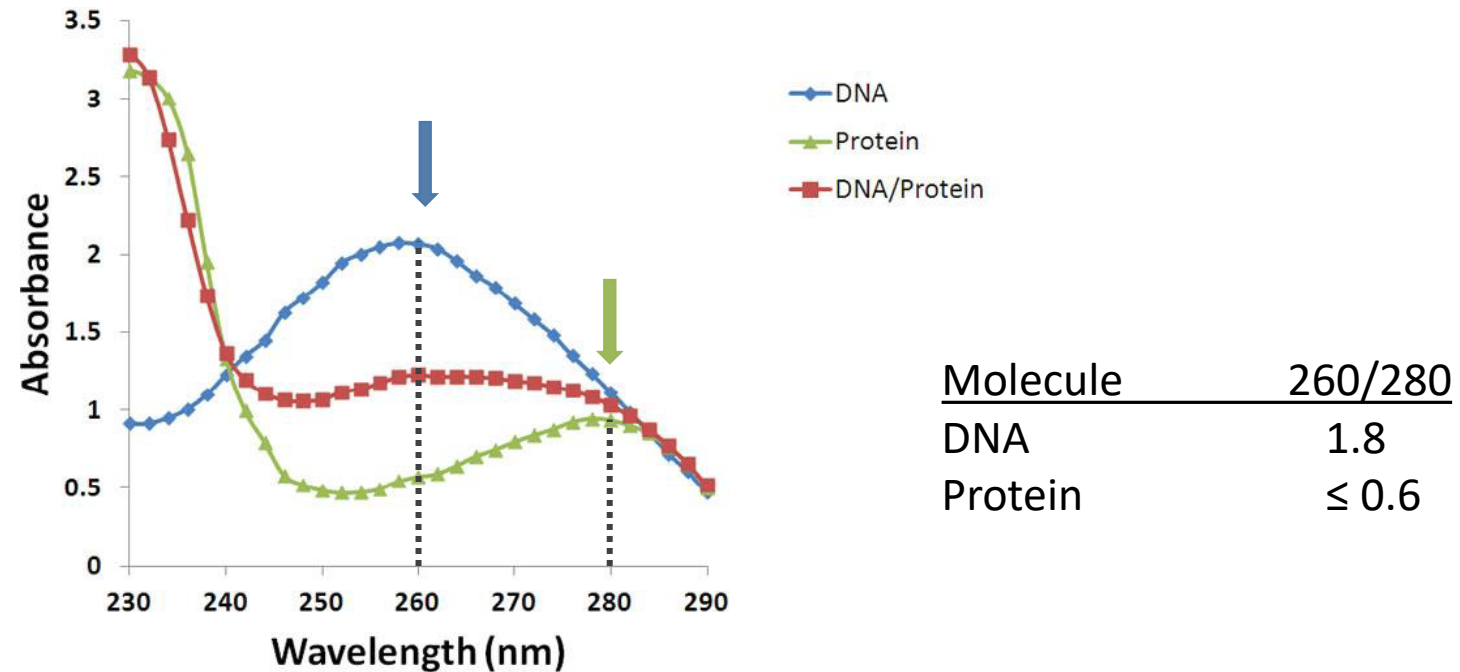
- UV spectrophotometry and the A260:280 ratio



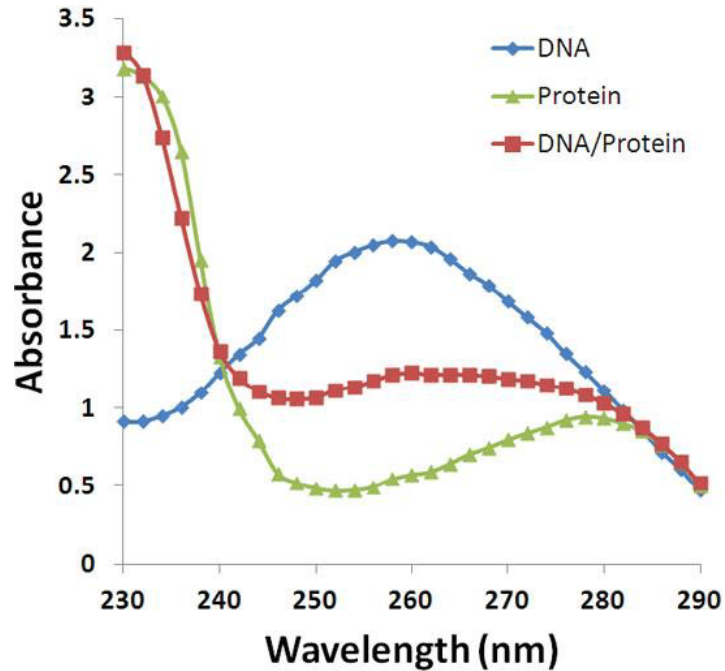
Analyse the UV spectrum of your sample

Estimating Purity Using UV Spectrophotometry

- UV spectrophotometry and the A260:280 ratio

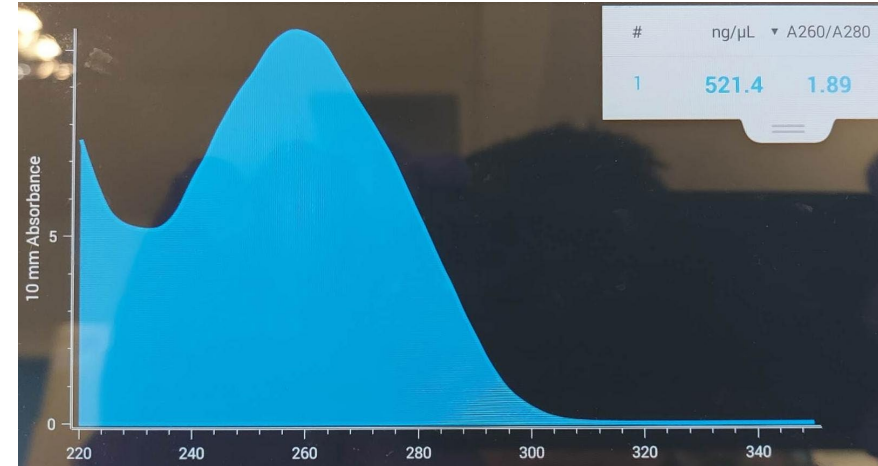


Estimating Purity Using UV Spectrophotometry

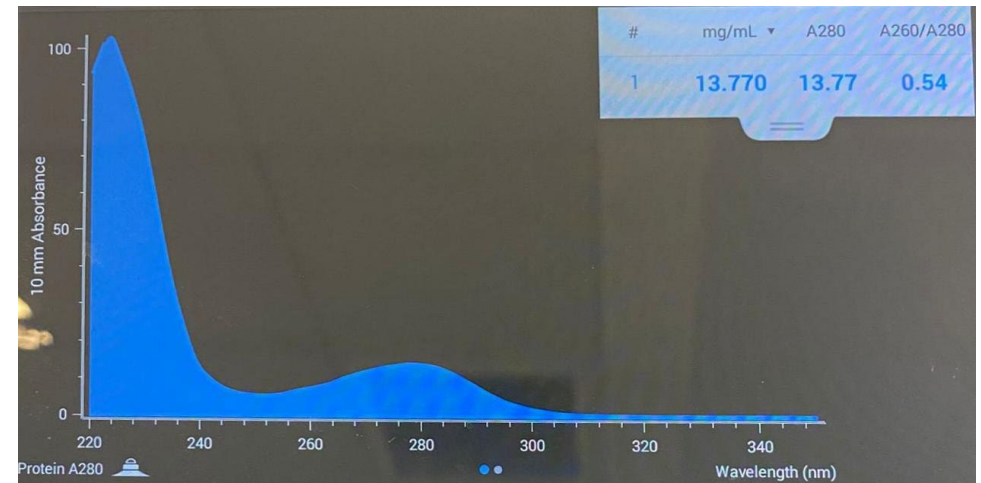


Molecule	260/280
DNA	1.8
Protein	≤ 0.6

Pure DNA
sample on
Nanodrop

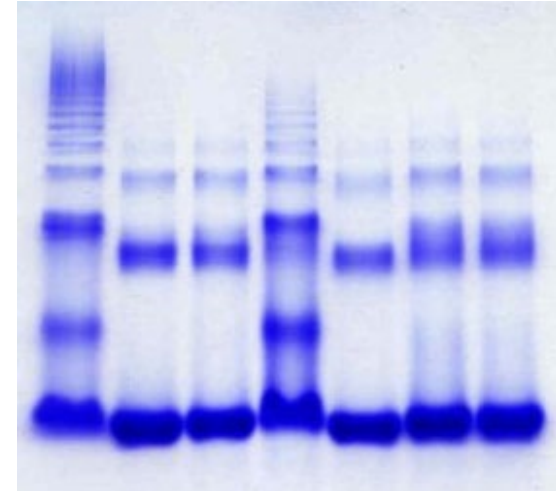
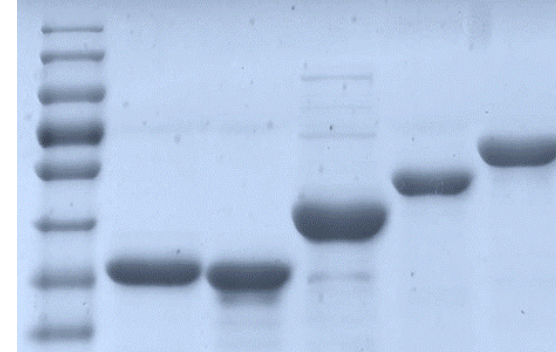


Pure protein
sample on
Nanodrop



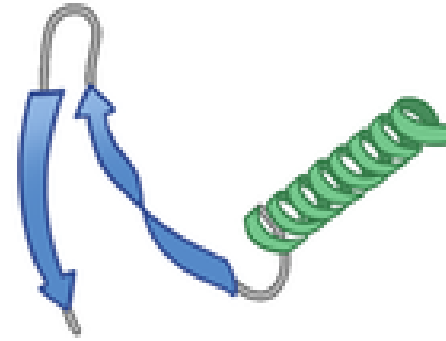
Is my protein folded?

- For normal SDS-PAGE the sample is boiled in SDS so is unfolded to allow for good size separation
- Native-PAGE doesn't unfold the protein but is mostly useful for monitoring interactions or oligomerisation



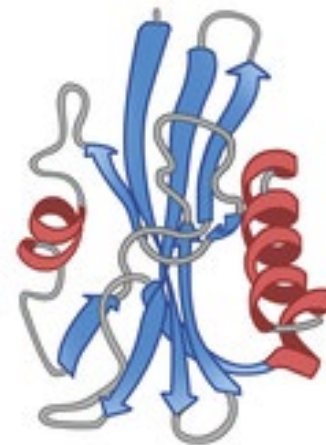
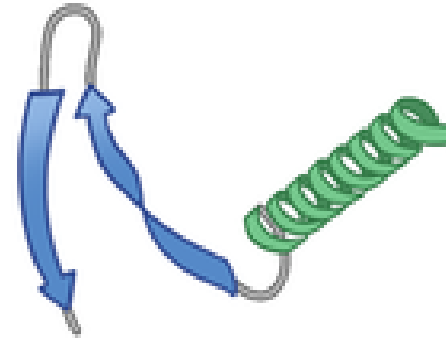
Is my protein folded?

- Secondary structure
 - Check this using circular dichroism



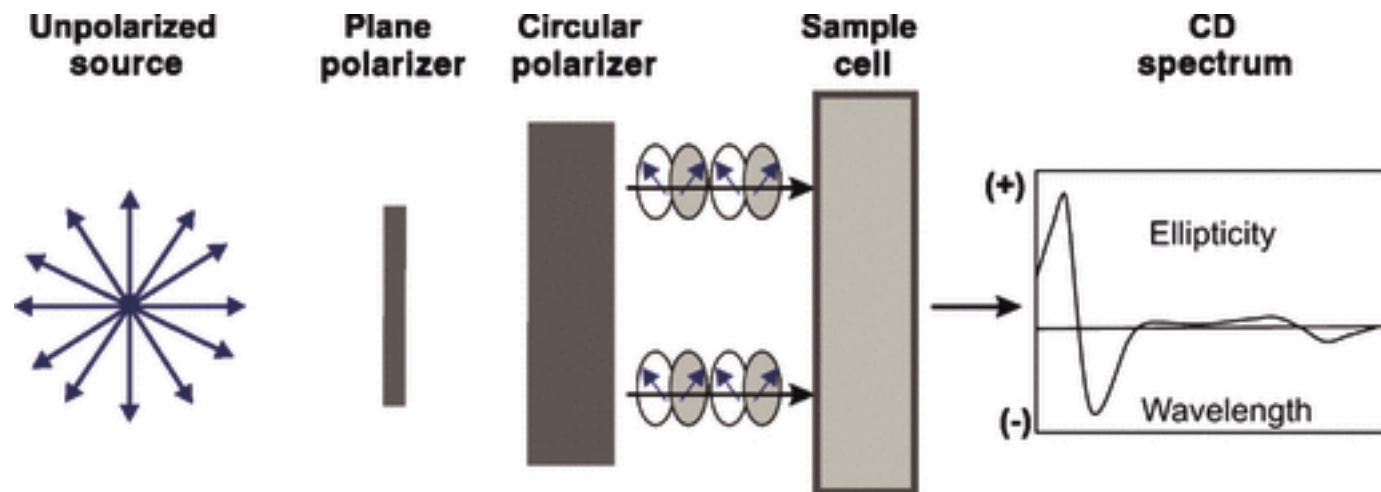
Is my protein folded?

- Secondary structure
 - Check this using circular dichroism
- Tertiary structure
 - Check this using differential scanning calorimetry



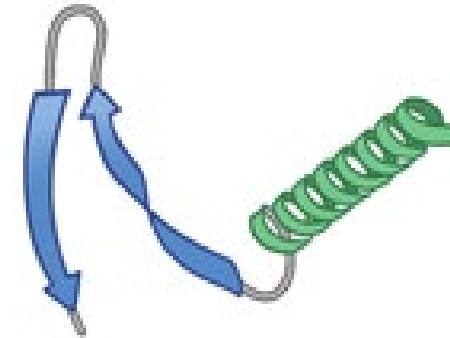
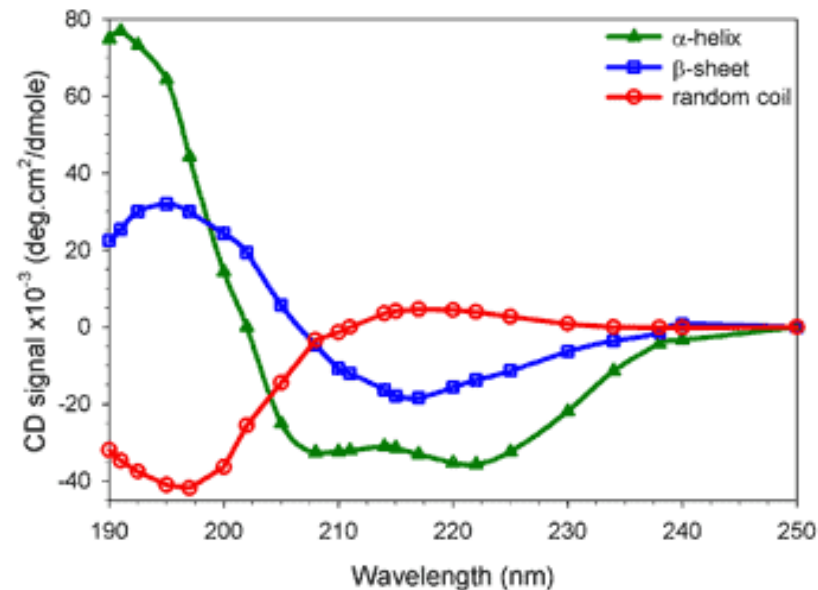
Circular dichroism (CD) spectroscopy

- Light can be circularly polarised (left- or right-handed)
- CD spectroscopy measures difference in absorption of left- and right-handed circularly polarised light



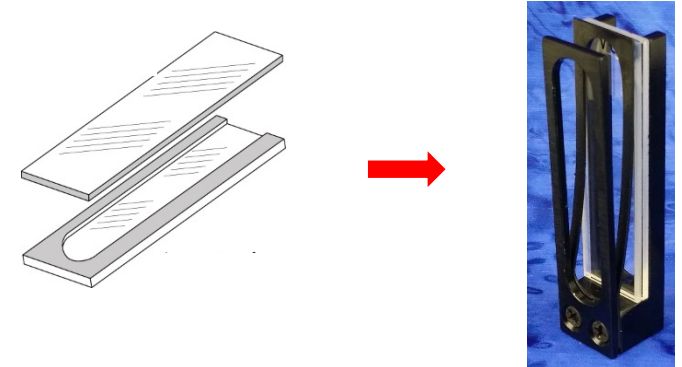
CD of proteins

- Protein secondary structural elements have characteristic CD spectra
- CD can tell you the secondary structure of your protein



Measuring CD spectra

- Put sample into thin quartz cuvette
 - Need buffer with low optical activity and low salt
- Measure spectra
- Tells you secondary structure, **not** folding
 - Although you can do 'melt curves'



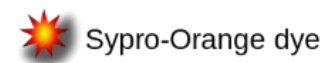
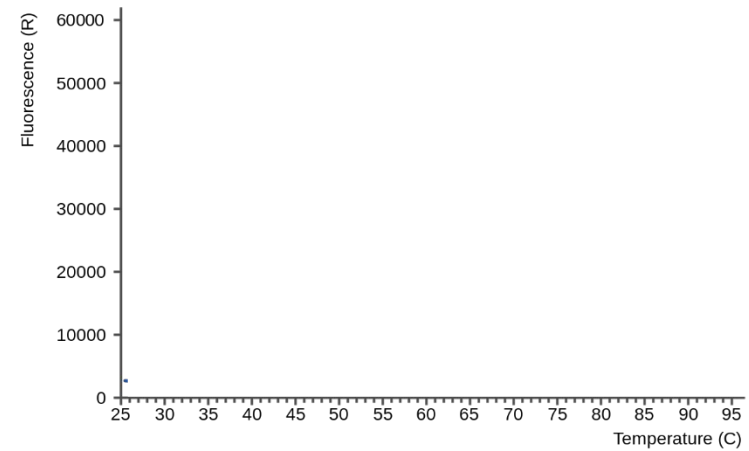
Measuring folding: Thermal Shift Assay

- Differential scanning fluorimetry
- Heat protein in presence of hydrophobic dye



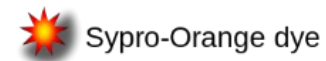
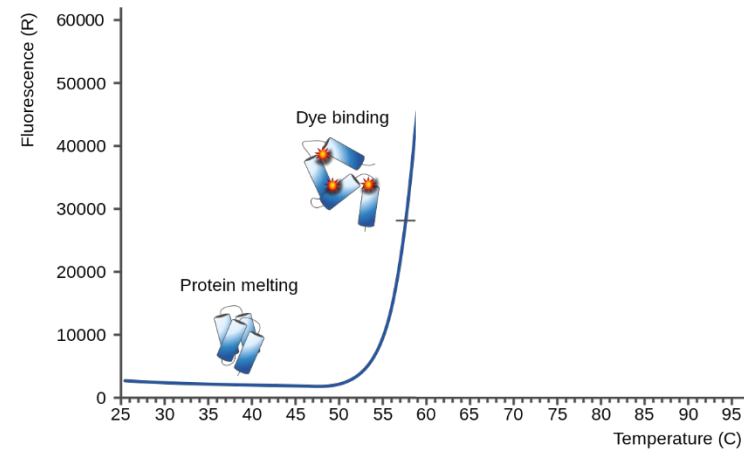
Measuring folding: Thermal Shift Assay

- Differential scanning fluorimetry
- Heat protein in presence of hydrophobic dye



Measuring folding: Thermal Shift Assay

- Differential scanning fluorimetry
- Heat protein in presence of hydrophobic dye
 - Fluorescence goes up when dye binds unfolded protein



Sypro-Orange dye

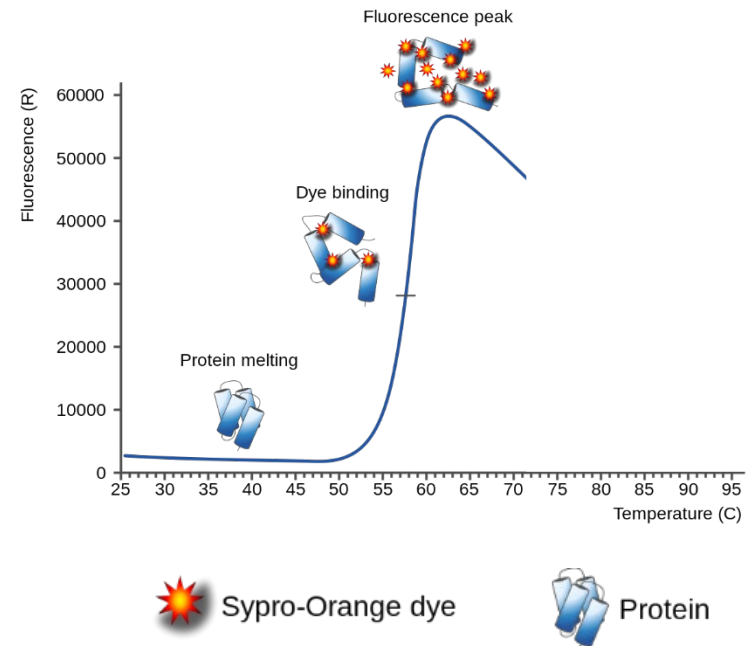


Protein



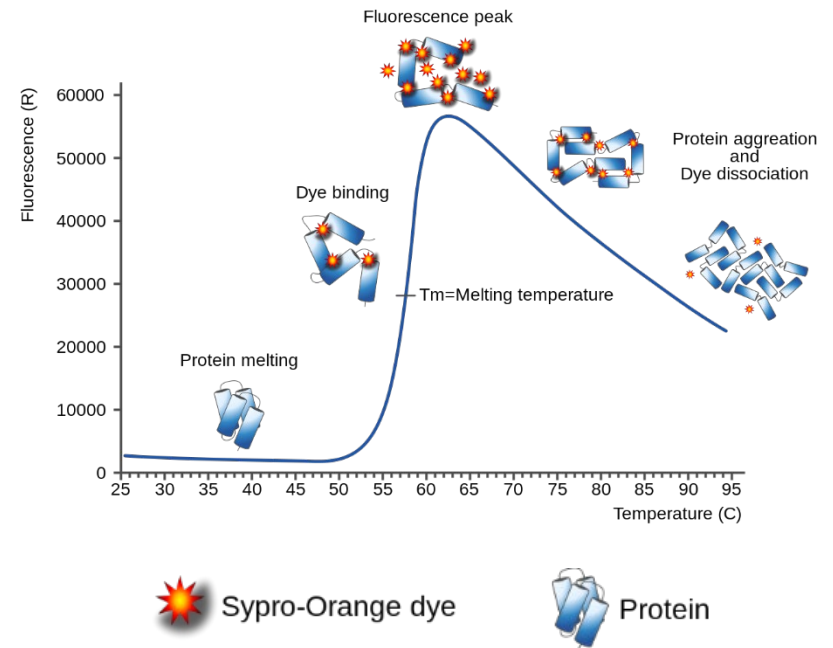
Measuring folding: Thermal Shift Assay

- Differential scanning fluorimetry
- Heat protein in presence of hydrophobic dye
 - Fluorescence goes up when dye binds unfolded protein



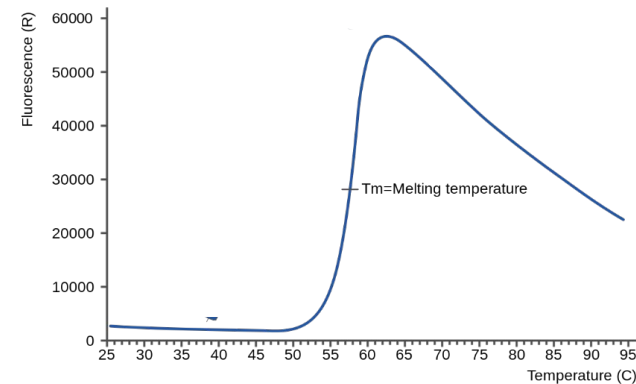
Measuring folding: Thermal Shift Assay

- Differential scanning fluorimetry
- Heat protein in presence of hydrophobic dye
 - Fluorescence goes up when dye binds unfolded protein



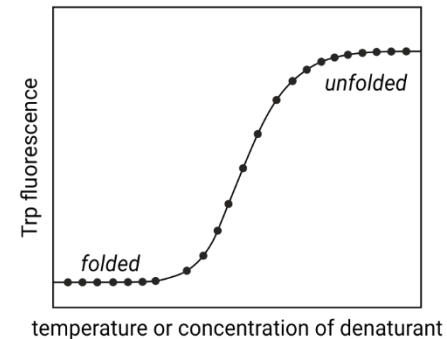
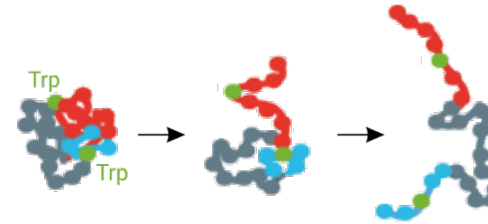
Measuring folding: Thermal Shift Assay

- Differential scanning fluorimetry
- Heat protein in presence of hydrophobic dye
 - Fluorescence goes up when dye binds unfolded protein
- Cheap and easy
 - qPCR machine



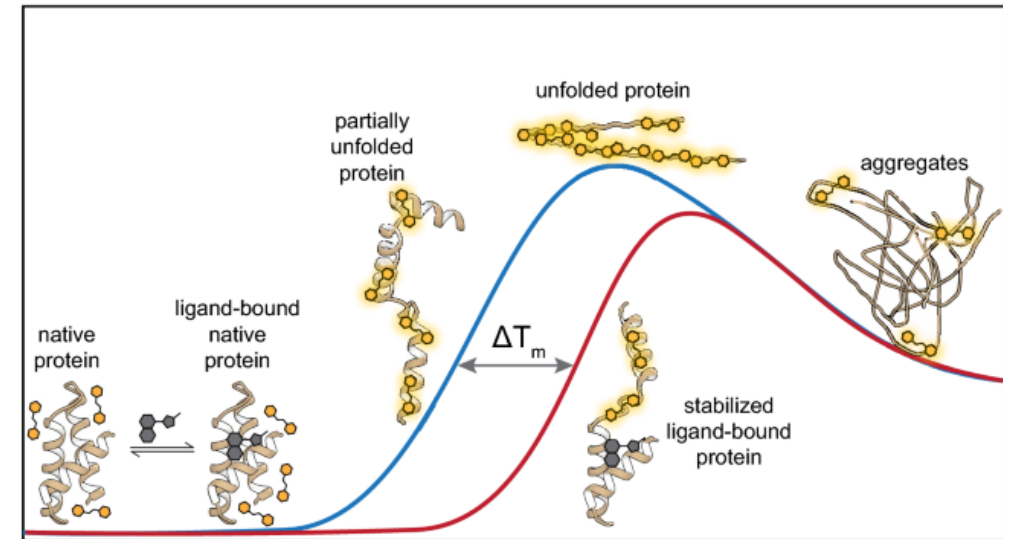
Measuring folding: Thermal Shift Assay

- Differential scanning fluorimetry
- Heat protein in presence of hydrophobic dye
 - Fluorescence goes up when dye binds unfolded protein
- Cheap and easy
 - qPCR machine
 - Can do label-free, Trp fluorescence



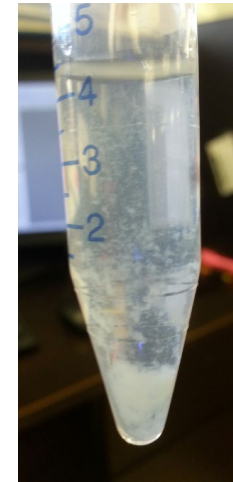
Why measure T_m ?

- Does your mutation disrupt the fold?
- What is the best buffer for my protein?
- Does your protein bind a ligand?
 - Ligand binding can stabilise proteins



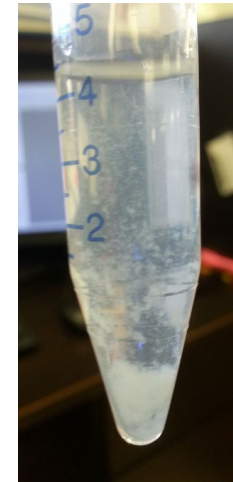
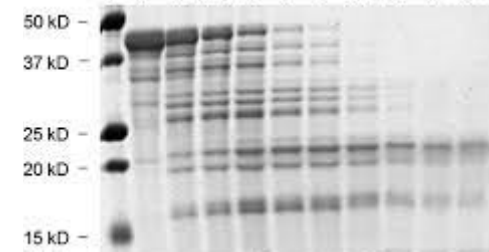
Protein Storage

- Purified proteins are not always stable for very long at 4°C
- Over time they can degrade or precipitate
- Proteins can also precipitate if they are at too high a concentration



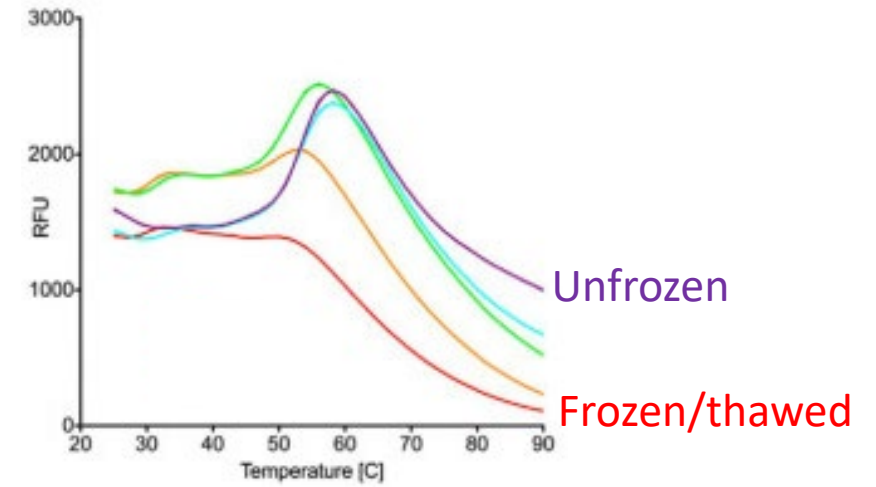
Protein Storage

- Ideally use your protein when it is fresh
- If you need to store it you can snap-freeze it in liquid nitrogen in small aliquots (50 μ L)
- Either freeze it at high conc > 2 mg/mL or add some glycerol (5-20%)
- However, some proteins don't freeze well...



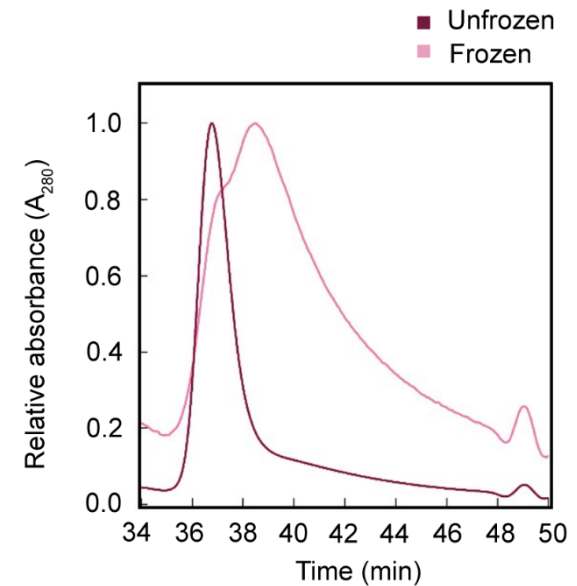
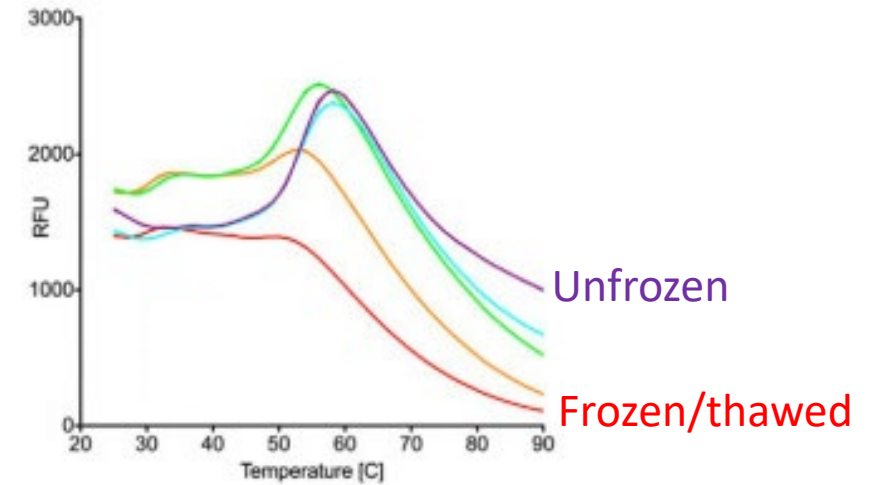
Checking quality of frozen protein

- Redo melt curve, DSF



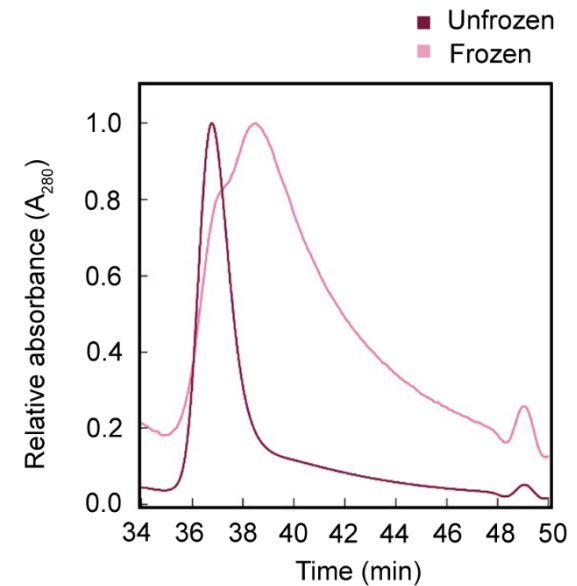
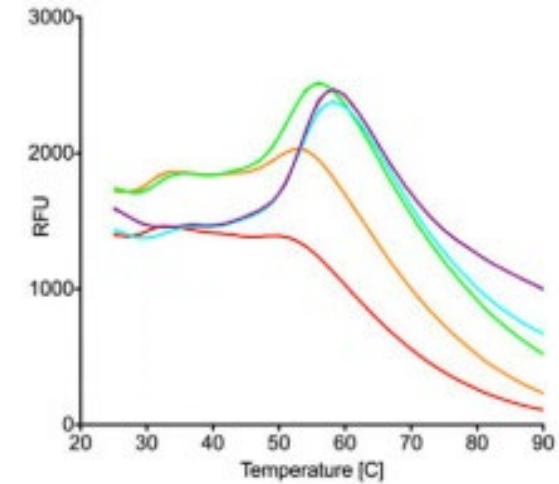
Checking quality of frozen protein

- Redo melt curve, DSF
- Run SEC again to see if there is an aggregate peak



Checking quality of frozen protein

- Redo melt curve, DSF
- Run SEC again to see if there is an aggregate peak
- Run an activity assay
- Do a pulldown experiment with a binding partner



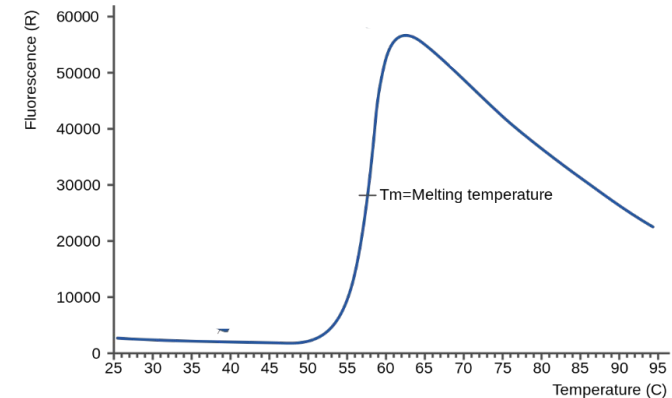
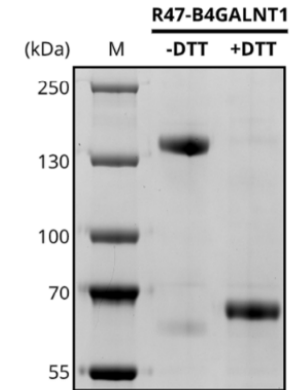
Troubleshooting and Optimisation

- Different buffers during purification
- Ensure your buffer is ≥ 1 pH unit from the pI of your protein
- Some proteins are more stable with addition of:
 - 5-10% glycerol
 - Increased [NaCl], up to 1M
 - Divalent cations: Ca, Mg
- Design a new construct!



Protein Purification Part 3 – Quality Control

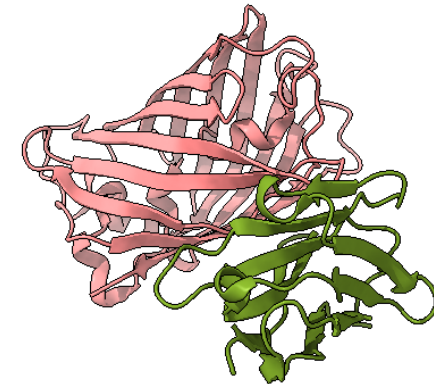
- Today we learnt about:
 - Interpreting your data
 - Checking your protein quality
 - Checking for correct protein folding
 - How to store your protein
 - Troubleshooting difficult samples



End of Week 1 Lectures

- We've learnt about:
 - Protein expression
 - Protein purification
 - Chromatography
 - Checking protein quality

- Next week we learn about:
 - Biophysics
 - Protein-protein interactions
 - Looking at protein structures
 - Solving and predicting protein structure



Questions?

Then let's do some SDS-PAGE

